# Photocyclisation of Enamides. Part 35. ${ }^{1.2}$ New Total Syntheses of the Ergot Alkaloids ( $\pm$ )-Chanoclavine-I and ( $\pm$ )-Isochanoclavine-I using a Fragmentation of 3-Amino Alcohols 

Ichiya Ninomiya,* Naoko Habe (née Kuninobu), Toshiko Kiguchi and Takeaki Naito<br>Kobe Women's College of Pharmacy, Motoyamakita, Higashinada, Kobe 658, Japan


#### Abstract

A new synthetic route involving the fragmentation reaction of 3-amino alcohols for the total synthesis of 6,7 -secoergoline alkaloids was developed and then successfully applied to the total syntheses of $( \pm)$-chanoclavine-I and ( $\pm$ )-isochanoclavine-I.


In continuation of our synthesis of ergot alkaloids ${ }^{3}$ which features a strategy of first building up the ergoline skeleton by enamide photocyclisation followed by elaboration of particular structural features, attention has been focussed on the application of a common skeletal compound to the synthesis of other members of this group of alkaloids, ${ }^{4}$ namely the ring-opened and modified components, thereby establishing a general synthetic methodology for all members of ergot alkaloids. We picked the fragmentation reaction of the 3 -amino alcohols, the structure commonly appearing in the photocyclised products, as the key reaction for the synthesis of ring-opened alkaloids of the chanoclavine type. ${ }^{4}$

Heterolytic fragmentation of 3-amino alcohol derivatives had been developed by Grob, ${ }^{5}$ who demonstrated that by base treatment the bicyclic mesate $\mathbf{A}$ underwent smooth fragmentation via either peripheral cleavage or internal cleavage ${ }^{6}$ to give the iminium compound $\mathbf{B}$, which was then hydrolysed to the cyclohexane derivative D (Scheme 1). Previous studies ${ }^{5}$ on this reaction have only been carried out on some simple 3 -amino alcohol systems, therefore leaving the reaction mechanism to be clarified. We first investigated the fragmentation of model compounds having a 1 -hydroxy-2-substituted octahydrobenzo $[f]$ quinoline structure and the results were then successfully applied to the total syntheses of $( \pm)$-chanoclavine-I and ( $\pm$ )-isochanoclavine-I.


Scheme 1 Reactions: i, peripheral cleavage; ii, internal cleavage; iii, aza-Cope rearrangement

## Results and Discussion

Preparation of 1-Hydroxyoctahydrobenzo[ f$]$ quinolines.-1-Hydroxy-1,2,3,4,4a,5,6,10b-octahydrobenzo[ $f$ ] ]quinolines were considered to be substrates suitable for preliminary investigation of the fragmentation. Seven benzo[ $f$ ]quinolines having two types of substituent (methyl and acetoxymethyl groups) at the 2 -position and different ring junctures ( $\mathrm{B} / \mathrm{C}$ ) were synthesized. The в/C-trans-2-monosubstituted derivatives 4-7,
which have one substituent at the 2 -position and a $\mathrm{B} / \mathrm{C}$-trans structure, had already been prepared from the lactam 2 obtained by reductive photocyclisation of the enamide $1 .{ }^{7}$ The B/C-cis-2-monosubstituted derivatives 9 and 11 were prepared from the photocyclised $\mathrm{B} / \mathrm{C}$-cis-lactam 3 according to the reported procedure ${ }^{7}$ as follows. Ozonolytic cleavage of the dihydrofuran ring in the cis-lactam 3 in methanol, followed by reduction with lithium aluminium hydride, afforded the amino diol 8 in $76 \%$ yield. Selective acetylation of the primary hydroxy group in the diol 8 was achieved by treatment with acetic anhydride in methylene dichloride in the presence of 4 -(dimethylamino)pyridine (DMAP) at $0^{\circ} \mathrm{C}$ to give the monoacetate 9 in $74 \%$ yield. When the diol 8 was treated with methanesulphonyl chloride (mesyl chloride) in pyridine at $0^{\circ} \mathrm{C}$, the monomesate 10 was prepared in $80 \%$ yield, which was then reduced with sodium borohydride in dimethyl sulfoxide (DMSO), ${ }^{8}$ to give the 2-methyl alcohol 11 in $62 \%$ yield. The $\mathrm{B} / \mathrm{C}-$ trans-2-disubstituted derivative 14 was prepared via a route involving methylation at the 3a-position of the photocyclised lactam 2 as follows. Lithium diisopropylamide (LDA) was added to a solution of the lactam 2 and methyl iodide in tetrahydrofuran (THF) at $-40^{\circ} \mathrm{C}$ to give the desired 3a-methyl lactam 12 in $87 \%$ yield. Inverse addition of methyl iodide to a solution of the lithiated lactam, prepared from compound 2 with LDA, gave the desired lactam 12 in only $54 \%$ yield, as well as formation of the ring-opened methylated lactam 13 in $2 \%$ yield. The stereochemistry of compound 12 as having a C/D-cis$11 \mathrm{~b}, 11 \mathrm{c}$-anti configuration was deduced from the similarity of its proton nuclear magnetic resonance ( ${ }^{1} \mathrm{H}$ NMR) spectrum with that of the starting lactam 2 and also by the observation of a nuclear Overhauser effect (NOE) $(24 \%)$ between the 3amethyl group and 11c-hydrogen. The by-product 13 was also characterised from its spectral data except on the relative configuration of the 4 a -hydrogen and the substituent at the 2 position. Ring opening of the dihydrofuran ring of the methyl lactam 12 by the two-step method involving ozonolysis and lithium aluminium hydride reduction gave the diol 14 in $79 \%$ yield. Selective acetylation of the diol 14 was performed by treatment with acetic anhydride in methylene dichloride in the presence of DMAP at $0^{\circ} \mathrm{C}$ to afford the monoacetate 15 in $99 \%$ yield.

The stereochemistry of the $\mathrm{B} / \mathrm{C}$-trans-amino alcohols 4-7, and 15 was established from their ${ }^{1} \mathrm{H}$ NMR spectra, particularly the coupling constants ( $10-11 \mathrm{~Hz}$ ) between the 1 - and $10 b$-proton and those $(10-11 \mathrm{~Hz})$ between the $4 a$ - and 10 b -proton. The results were consistent with a stable chair conformation for ring C with the 1-hydroxy group in an equatorial orientation, as shown by the structure $\mathbf{E}$. On the other hand, the axial orientation of the 1 -hydroxy group in the $\mathrm{B} / \mathrm{C}$-cis-amino alcohols 9 and 11 was deduced from the coupling constants (3

Table 1 Fragmentation of the 3-amino alcohol derivatives 4, 5 and 16-20

| Substrate | Reaction conditions | Yield (\%) of products |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | 22 or 23 | 24 or 26 | 25 or 27 | 29 | 30 |
| 4 | MsCl in pyridine, $0^{\circ} \mathrm{C}, 3.5 \mathrm{~h}$ | 34 |  |  |  |  |
| 4 | $\mathrm{MsCl}, \mathrm{Et}_{3} \mathrm{~N}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}, 0^{\circ} \mathrm{C}, 5 \mathrm{~h}$ | 39 |  |  |  |  |
| 4 | $\mathrm{MsCl}, \mathrm{Et}_{3} \mathrm{~N}$ in toluene, $0^{\circ} \mathrm{C}, 2.5 \mathrm{~h}$ | 54 |  |  |  |  |
| 5 | $\mathrm{MsCl}, \mathrm{Et}_{3} \mathrm{~N}$ in toluene, $0^{\circ} \mathrm{C}, 2 \mathrm{~h}$ | 44 |  |  |  |  |
| 16 | $80 \% \mathrm{EtOH}, 20^{\circ} \mathrm{C}, 4$ days | 30 | Trace | 20 |  |  |
| 16 | $80 \% \mathrm{EtOH}, 50^{\circ} \mathrm{C}, 6 \mathrm{~h}$ | 29 | 9 | 29 |  |  |
| 16 | $80^{\%} \mathrm{EtOH}, 70^{\circ} \mathrm{C}, 6 \mathrm{~h}$ | 39 | Trace | 44 |  |  |
| 16 | Absolute EtOH, $50{ }^{\circ} \mathrm{C}, 6 \mathrm{~h}$ | 11 | 16 | 44 |  |  |
| 16 | Neat, $110^{\circ} \mathrm{C}, 7 \mathrm{~h}$ | Trace | 24 | Trace |  |  |
| 17 | $80 \% \mathrm{EtOH}, 70^{\circ} \mathrm{C}, 4 \mathrm{~h}$ | 27 | 20 | 31 |  |  |
| 17 | Neat, $110^{\circ} \mathrm{C}, 5 \mathrm{~h}$ | Trace | 33 | Trace |  |  |
| 18 | $80 \% \mathrm{EtOH}, 70{ }^{\circ} \mathrm{C}, 29 \mathrm{~h}$ | 30 |  | 27 |  |  |
| 18 | Neat, $110^{\circ} \mathrm{C}, 13 \mathrm{~h}$ | Trace |  |  |  |  |
| 19 | $80 \% \mathrm{EtOH}, 70^{\circ} \mathrm{C}, 26 \mathrm{~h}$ | 25 |  | 30 |  |  |
| 20 | $80 \% \mathrm{EtOH}, 20^{\circ} \mathrm{C}, 22 \mathrm{~h}$ |  |  |  |  | 6 |
| 20 | $80 \% \mathrm{EtOH}, 70^{\circ} \mathrm{C}, 2 \mathrm{~h}$ |  |  |  | 50 | 25 |
| 20 | Absolute EtOH, $50{ }^{\circ} \mathrm{C}, 4 \mathrm{~h}$ |  |  |  | 79 | 6 |
| 20 | Absolute EtOH, $70^{\circ} \mathrm{C}, 4 \mathrm{~h}$ |  |  |  | 73 | 11 |
| 20 | Neat, $110^{\circ} \mathrm{C}, 3.5 \mathrm{~h}$ |  |  |  | 39 |  |



1

$4 \mathrm{R}=\beta-\mathrm{CH}_{2} \mathrm{OAC}$
$5 \mathrm{R}=\beta-\mathrm{Me}$
$6 \mathrm{R}=\alpha-\mathrm{CH}_{2} \mathrm{OAc}$
$7 R=\alpha-M e$

$\mathrm{R}=\mathrm{CH}_{2} \mathrm{OH}$
R $=\mathrm{CH}_{2} \mathrm{OAC}$
$\mathrm{R}=\mathrm{CH}_{2} \mathrm{OMs}$
$\mathrm{R}=\mathrm{Me}$


12

$14 \mathrm{R}=\mathrm{H}$
$15 R=A c$


E (B/C-trans)


F (B/C-cis)

Hz ) between the 1 - and 10b-proton in their ${ }^{1} \mathrm{H}$ NMR spectra, as shown by the conformation $F$. Thus we have prepared seven benzo[ $f$ ]quinolines, compounds 4-7,9,11 and 15, and the
fragmentation of these compounds was investigated as described below.

Fragmentation Reaction of Benzo[ f$]$ quinolines.-For facilitating ring opening, five amino alcohols, compounds 6, 7, 9, 11 and 15, were first treated with mesyl chloride in pyridine or toluene in the presence of triethylamine to give the corresponding mesates $\mathbf{1 6 - 2 0}$ in good yield. Under the same mesylating conditions, the $\mathrm{B} / \mathrm{C}$-trans-amino alcohols 4 and 5 with a 2 -equatorial substituent gave no isolable mesyl derivative, and instead underwent smooth fragmentation as shown in Table 1. The 2 -acetoxymethyl-1-ol 4 gave the $E$-amine 22 in $34-54 \%$ yield, together with the chloride 21. The 2-methyl1 -ol 5 gave the $E$-amine 23 in $44 \%$ yield under the same mesylating conditions. The structures of these products 22 and 23 were characterised as having an $E$-double bond and a methylamino group in a trans configuration on the tetrahydronaphthalene ring from their mass, IR and ${ }^{1} \mathrm{H}$ NMR spectra, particularly the coupling constants ( $15 \mathrm{~Hz} \mathrm{)} \mathrm{between}$ two olefinic protons and those ( $8-9 \mathrm{~Hz}$ ) between the 1 - and 2-proton in their ${ }^{1} \mathrm{H}$ NMR signals.
The isolated mesate $16^{7 b}$ was then subjected to fragmentation under various conditions and the results obtained are summarised in Table 1. Stirring of an $80 \%$ ethanolic solution of the mesate 16 at $20^{\circ} \mathrm{C}$ for 4 days gave a mixture of the $E$-trans- and $E$-cis-amine 22 and 25 in 30 and $20 \%$ isolated yield, respectively, together with a trace amount of the Z-transamine 24. Heating of the mesate 16 in either $80 \%$ ethanol or absolute ethanol at $50-70^{\circ} \mathrm{C}$ for 6 h afforded three amines, compounds 22, 24 and 25 , in $67-83 \%$ combined yields as shown in Table 1. Heating of the mesylester 16 without solvent at $110^{\circ} \mathrm{C}$ for 7 h gave the $Z$-amine 24 as the major product in $24 \%$ yield. The fragmentation of the 2 -methyl- $\mathrm{B} / \mathrm{C}$-trans-mesate $17^{76}$ proceeded smoothly to give the amines 23,26 and 27 in the yields shown in Table 1. The structures of the $Z$-transamines $\mathbf{2 4}$ and $\mathbf{2 6}$ were established from the coupling constants $(10-11 \mathrm{~Hz})$ between the two olefinic protons and those ( $9-10$ Hz ) between the 1 - and 2 -proton in their ${ }^{1} \mathrm{H}$ NMR spectra. The structure of the $E$-cis-amines 25 and 27 was deduced from the coupling constants ( $15-16 \mathrm{~Hz}$ ) between the two olefinic protons and those ( $5-6 \mathrm{~Hz}$ ) between the 1 - and 2-proton in their ${ }^{1} \mathrm{H}$ NMR spectra. Similarly, the fragmentation of the $\mathbf{B} / \mathrm{C}$-cismesates 18 and 19 was carried out in $80 \%$ ethanol or without solvent and the stereoisomers 22 and 25 from 18, and 23 and 27
from 19, were obtained in the yields shown in Table 1, though the consumption of the starting compounds 18 and 19 required a longer reaction time as compared with the $\mathrm{B} / \mathrm{C}$-trans-mesates 16 and 17. Marked differences between the $2 \beta$-substituted amino alcohols 4 and 5 and the $2 x$-substituted amino alcohols $6,7,9$ and 11 in their mesylation properties and also in the reactivity of the isolated mesylesters $16-19$ in the fragmentation can be explained as follows. As Grob ${ }^{5}$ has suggested that the fragmentation of the 3 -amino alcohol derivates requires an extended, anti-periplanar relationship between the leaving group, the $C-C$ bond which is undergoing cleavage, and the nitrogen lone pair electrons, ${ }^{5}$ the 1,2-trans-amino alcohols 4 and 5 can exist preferentially in the conformation $\mathbf{G}$ with an $x$ axial $N$-methyl group and $\beta$-equatorial lone-pair electrons; ${ }^{9}$ therefore the fragmentation of these amino alcohols 4 and 5 would proceed readily, even under the mesylating conditions, via the presumably formed mesate in situ. On the other hand, there exists considerable steric repulsion between the $2 \alpha$-axial substituent and the $\alpha$-axial methyl group on nitrogen in the conformation $G$ of the 1,2-cis-mesates 16 and 17. The ${ }^{1} \mathrm{H}$ NMR spectra of the $B / C$-cis-mesylesters 18 and 19 suggest that the mesyloxy group at the 1-position would be in an axial orientation as shown in the conformation $\mathbf{F}$, which would be unfavourable to fragmentation, thus retarding the fragmentation of $\mathrm{B} / \mathrm{C}$-cis-mesates 18 and 19 compared with that of the B/C-trans-mesates 16 and 17 which have an equatorial mesyloxy group at the 1-position.

Next, in order to establish the reaction pathway of the fragmentation, we investigated the reaction of compounds 16 and 18 in the presence of sodium borohydride. The mesates 16 and 18 were treated with sodium borohydride in $80 \%$ ethanol, and reacetylation of the crude products gave the identical benzazecine $\mathbf{2 8}$ in $55-79 \%$ yield. The structure of the product $\mathbf{2 8}$ was easily established from its spectral data, particularly by the ${ }^{1} \mathrm{H}$ NMR signals of olefinic protons at $\delta 6.90(\mathrm{~d}, J 16 \mathrm{~Hz}$, $8-\mathrm{H})$ and 5.26 (dd, $J 16,9 \mathrm{~Hz}, 7-\mathrm{H}$ ). This finding, and the result that a mixture of stereoisomers was obtained by fragmentation of the mesates 16-19, indicated that both the ring-opened iminium I (formed via peripheral cleavage) and the azecine H (formed via internal cleavage) would exist in an equilibrium as intermediates in the fragmentation. Since the ten-membered intermediates $\mathbf{H}$ would possibly exist in various conformations due to flexibility, we assumed that the yields and ratios of products $22-27$ were greatly influenced by subtle difference in the reaction conditions employed.

When the B/C-trans-2-disubstituted mesylester 20 was subjected to the fragmentation as shown in Table 1, the $Z$-amine 29 was obtained as a major product, in addition to the $E$-amine 30 as a minor product. The relative configurations of the two substituents on the tetrahydronaphthalene moiety of both products 29 and 30 were established to be trans by the coupling constants ( 10 Hz ) between the 1 - and 2-protons of their ${ }^{1} \mathrm{H}$ NMR spectra. The configurations of olefinic double bonds were suggested to be $Z$ in compound 29 and $E$ in compound 30 , by comparison of the signal pattern due to the allylic methylene protons [ $\delta 4.86$ and $4.74(\mathrm{ABq}, J 13 \mathrm{~Hz})$, in $29 ; \delta 4.62(\mathrm{~s})$ in 30] in their ${ }^{1} \mathrm{H}$ NMR spectra with that of the natural alkaloids, isochanoclavine-I and chanoclavine-I. Oppolzer ${ }^{10}$ and Stauffacher ${ }^{11}$ have independently reported that isochanoclavine-I and chanoclavine-I exhibited ${ }^{1} \mathrm{H}$ NMR signals of allylic methylene protons at $\delta \sim 4.6$ as an AB quartet and $\delta 4.4$ as a singlet, respectively. The major product 29 has the stereochemistry readily suggested from the reaction course on the direct cleavage between the 2 - and 3-position in the fragmentation compound of $\mathbf{2 0}$. However, the minor product $\mathbf{3 0}$ has the stereochemistry suggested from the reaction course via the azecine derivative $H$. Further treatment of the $\mathrm{B} / \mathrm{C}$-trans-2disubstituted mesate 20 with sodium borohydride in $80 \%$
ethanol, followed by reacetylation, gave the benzazecine 31 in $74 \%$ yield. The structure of compound 31 was deduced from its ${ }^{1} \mathrm{H}$ NMR spectrum $[\delta 6.81(\mathrm{~d}, J 16.5 \mathrm{~Hz}, 8-\mathrm{H})$ and $5.30(\mathrm{~d}, J$ $16.5 \mathrm{~Hz}, 7-\mathrm{H})$ ].

Thus, we have developed a new and simple synthetic method for 6,7 -secoergolines using the fragmentation reaction of the 3 -amino alcohol derivatives, particularly giving the expected compounds 22, 23 and 29 as major products from the amino alcohols 4,5 and 20 having the $2 \beta$-substituent and b/C-transring junction, respectively.

$6 \mathrm{R}^{1}=\mathrm{H}, \mathrm{R}^{2}=\mathrm{CH}_{2} \mathrm{OAC}, \mathrm{R}^{3}=\beta-\mathrm{H}$ $17 R^{1}=H, R^{2}=\mathrm{Me}, R^{3}=\beta-H$
$\mathrm{R}^{1}=\mathrm{H}, \mathrm{R}^{2}=\mathrm{CH}_{2} \mathrm{OAC}, \mathrm{R}^{3}=\alpha-\mathrm{H}$


21

$$
\begin{aligned}
& R^{1}=H, R^{2}=M e, R^{3}=\alpha-H \\
& R^{1}=M e, R^{2}=C H_{2} O A C, R^{3}=\beta-H
\end{aligned}
$$



Total Synthesis of $( \pm)$-Chanoclavine-I and ( $\pm$ )-Isochano-clavine-I.-According to the results on the model compounds shown in the previous chapter, the total synthesis of 6,7secoergolines was carried out by applying fragmentation of the amino alcohols 38 and 40 which are readily available from the known ${ }^{3 a}$ photocyclised lactams 33 and 34 and are different in respect of their stereochemistry at the $C / D$ ring junction.

Two amino alcohols, compounds 38 and 40, were synthesized from the known ${ }^{3 a}$ key intermediates 33 and 34 which had been obtained by reductive photocyclisation of the enamide $32^{3 a}$ and successfully used as the common intermediates in the total synthesis of ergoline-type alkaloids. ${ }^{3}$ The C/D-trans-lactam $\mathbf{3 3}^{3 a}$ was methylated with methyl iodide and LDA at $-40^{\circ} \mathrm{C}$ in THF as described above, to give the desired 3a-methyl lactam 35 in $89 \%$ yield as the sole product. The stereochemistry of compound 35 as having the $\mathrm{D} / \mathrm{E}$-cis- $11 \mathrm{~b}, 11 \mathrm{c}$-anti configuration was deduced from the similarity of its ${ }^{1} \mathrm{H}$ NMR spectrum with that of the starting lactam 33 and also by the observation of NOE ( $25 \%$ ) between the 3a-methyl group and the 11c-

Table 2 Fragmentation of the mesylesters 42 and 47

|  |  |  |  |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |



32

$33 \mathrm{R}=\mathrm{H}$
$35 \mathrm{R}=\mathrm{Me}$

$34 \mathrm{R}=\mathrm{H}$ $39 \mathrm{R}=\mathrm{Me}$

$36 R^{1}=R^{2}=H$ $37 \mathrm{R}^{1}=\mathrm{CH}_{2} \mathrm{Ph}, \mathrm{R}^{2}=\mathrm{H}$ $38 R^{1}=R^{2}=A c$


hydrogen. Ring opening of the dihydrofuran ring of the methyl lactam 35 by the two-step method ${ }^{3 b}$ involving ozonolysis and lithium aluminium hydride reduction gave the debenzoylated amino diol 36 in $53 \%$ yield together with the $N$-benzyl derivative 37 in $6 \%$ yield. Selective acetylation of the primary hydroxy and secondary amino groups in compound $\mathbf{3 6}$ was performed by treatment with acetic anhydride in pyridine at $0^{\circ} \mathrm{C}$ to afford the desired amino alcohol 38 in $93 \%$ yield. The equatorial orientation of the 9 -hydroxy group on the acetate 38 was suggested from the coupling constants ( 11 Hz ) between the 9 and 10 -proton in its ${ }^{1} \mathrm{H}$ NMR spectrum. In a similar manner, the known C/D-cis-lactam $34{ }^{3 a}$ was converted into the desired amino alcohol 40 via the 3a-methyl lactam 39 in $67 \%$ overall yield. The 9 -hydroxy group of the acetate $\mathbf{4 0}$ was suggested to have an equatorial orientation upon consideration of the coupling constants ( 11 Hz ) between the 9 - and 10 -proton in its ${ }^{1} \mathrm{H}$ NMR spectrum. We then thoroughly investigated the fragmentation of these substrates via their 9 -mesates.
Treatment of the acetate 38 with one mol equiv. amount of mesyl chloride led to only recovery of the starting compound 38, while similar treatment with an excess of mesyl chloride gave an intractable mixture. We then investigated direct formation of the desired allyl acetate $\mathbf{4 3}$ from the acetate $\mathbf{3 8}$ by fragmentation via the presumed mesate intermediate 42. The acetate 38 was treated with mesyl chloride ( 20 mol equiv.) in pyridine at $20^{\circ} \mathrm{C}$ for 3 h and the mixture was then warmed at $50^{\circ} \mathrm{C}$ for 4 h to give the desired fragmentation products 43 , 44 and 45 in 11, 43 and $5 \%$ isolated yield, respectively. Since the major product 44 would be formed by sulphonation of the desired product $\mathbf{4 3}$ with the excess of mesyl chloride used, we
investigated several reaction conditions where diethylamine was used as a quencher of the unchanged mesyl chloride as shown in Table 2. Best results were obtained when, after addition of diethylamine to a solution of the mesylated reaction mixture at $-15^{\circ} \mathrm{C}$, the mixture was diluted with anhydrous ethanol and heated at $50^{\circ} \mathrm{C}$ for 3 h . Thereby the desired allyl acetate $\mathbf{4 3}$ was isolated in $56 \%$ yield. The stereochemistry of the products 43 and its isomer 45 was deduced by comparison of their ${ }^{1} \mathrm{H}$ NMR spectra with those of natural isochanoclavine-I and chanoclavine-II. ${ }^{10.11}$ The structure of the minor product 45 was deduced as the 5,10 -cis- - -allyl acetate judging from the coupling constant ( 6 Hz ) between the 5 - and 10 -proton and the signal pattern (singlet) of allylic methylene protons. The product 44 was identical with the sample which was prepared by mesylation of the product 43. Furthermore, treatment of the reaction mixture, which was obtained from the acetate 38, mesyl chloride, pyridine and diethylamine at $20^{\circ} \mathrm{C}$, with sodium borohydride in anhydrous ethanol followed by reacetylation gave the 10 -membered-ring amine $\mathbf{4 6}$ in $59 \%$ yield.

Next we investigated the fragmentation reaction of the c/D-cis-acetate 40. Mesylation of the $\mathrm{c} / \mathrm{D}$-cis-acetate 40 in pyridine gave the corresponding unstable mesyl derivative 47 which was without purification used in the next fragmentation reaction. According to the procedure exploited for the benzo[ $f$ ]quinolines, the mesate 47 was treated in absolute ethanol at $50^{\circ} \mathrm{C}$ for 3.5 h to give two products, 48 and 49 , in 11 and $22 \%$ yield, respectively. Structure of the minor product 48 was deduced to be the 5,10-trans- $E$-allyl acetate judging from the signal pattern of $10-\mathrm{H}$ which appeared as a triplet ( $J 10 \mathrm{~Hz}$ ) and that of the allylic methylene protons which appeared as a



Scheme 2 Reaction: i, aza-Cope rearrangement
singlet. Similarly, the structure of the product 49 was deduced to be the 5,10 -cis- $Z$-allyl acetate judging from the coupling constant $(6 \mathrm{~Hz})$ of the 5 - and 10 -proton and the signal pattern (AB quartet) of the allylic methylene protons. ${ }^{12}$ Interestingly, yields and ratios of the products 48 and 49 were not influenced by the solvent used but by the reaction temperature as shown in Table 2. The desired allyl acetate 48 for the synthesis of chanoclavine-I was obtained in $36 \%$ isolated yield by warming a solution of the mesyl compound 47 in ethylene glycol at $120^{\circ} \mathrm{C}$. The relationship between the substrates 38 and 40 and their products ( 43 and 45 ) and ( 48 and 49 ) is summarised as follows. ( $/ \mathrm{D}$-trans-Z-Product 43 has the expected structure from the proposed mechanism of the the fragmentation of the starting substrate 38 . Similarly, the allyl acetate 49 is an expected product from compound 47 . On the other hand, the allyl acetates 45 and 48 have the unexpected structures from the starting substrates $\mathbf{3 8}$ and $\mathbf{4 0}$ following consideration of the stereochemistry of C/D-ring and the relationships between the 9 -hydrogen and the 8 -methyl group. This result can be explained as follows. The allyl acetates 43 and 49 would be obtained by hydrolysis of the intermediate $\mathbf{J}$, while the allyl acetates $\mathbf{4 5}$ and 48 would be produced from the intermediate $L$ via the conformational conversion of $\mathbf{K}$ to boat-like intermediate $\mathbf{K}^{\prime}$, followed by aza-Cope rearrangement (Scheme 2). The requirement of a higher temperature for the formation of compound 48 can be reasonably explained by the proposed mechanism involving conformational inversion of $\mathbf{K}$ to $\mathbf{K}^{\prime}$. Results obtained in the previously described fragmentation of the disubstituted benzo[ $f$ ]quinoline 20 can be reasonably explained by conformational exchange of the intermediate $\mathbf{K}$ to $\mathbf{K}^{\prime \prime}$ as follows. The ten-membered-ring intermediate $\mathbf{K}$, formed from compound 20, would be transformed into $\mathbf{K}^{\prime \prime}$ via $\mathbf{K}^{\prime}$ by the rotation of single bonds around $\mathrm{C}=\mathrm{C}$ and $\mathrm{C}=\mathrm{N}$ double bonds. Additionally, since 3,5 -syn-products are known to be more stable in the ergot group of alkaloids, ${ }^{3 a .13}$ the number of the products formed by the fragmentation of the two substrates 42 and 47 would be limited to two ( 43 and 45 from 42; 48 and 49 from 47) in each case.

Finally, total synthesis of the natural alkaloids was completed by the conversion of the allyl acetates 43 and 48 into the respective target molecules. Hydrolysis of the acetates 43 and 48 by heating in methanol containing conc. hydrochloric acid at $80{ }^{\circ} \mathrm{C}$ followed by dehydrogenation ${ }^{3 b}$ with benzeneseleninic anhydride in the presence of 3 mol equiv. of indole and 10 mol equiv. of triethylamine in THF afforded the indole derivatives $50(83 \%$ yield from 43$)$ and 51 ( $67 \%$ yield from 48 ). The ${ }^{1} \mathrm{H}$

(土)-chanoclavine-I

NMR and IR spectra of the synthetic compounds $\mathbf{5 0}$ and $\mathbf{5 1}$ were found to be completely identical with those of authentic samples ${ }^{14}$ of $( \pm)$-isochanoclavine-I and ( $\pm$ )-chanoclavine-I, respectively, provided by Professor Somei. Thus, we have succeeded in the total synthesis of $( \pm)$-isochanoclavine-I and $( \pm)$-chanoclavine-I via a fragmentation of 3-amino alcohol derivatives.

## Experimental

${ }^{1} \mathrm{H}$ NMR spectra were measured with JEOL PMX-60 (60 MHz ), Varian XL-200 ( 200 MHz ), and VXR-500 ( 500 MHz ) instruments for solutions in deuteriochloroform unless otherwise stated (tetramethylsilane as internal reference); $J$-values are given in Hz . IR spectra were measured with a Hitachi 215 machine for solutions in chloroform unless otherwise stated. Mass spectra were taken with a Hitachi M-80 spectrometer. M.p.s were determined with a Kofler-type hot-stage apparatus and are uncorrected. Reactions were performed under nitrogen. Extracts from the reaction mixtures were washed with water, dried over anhydrous sodium sulfate, and evaporated under reduced pressure. TLC was performed on precoated silica gel 60F-254 ( 0.25 mm thick, Merck) and preparative TLC (PLC) on precoated silica gel $60 \mathrm{~F}-254$ ( 0.5 mm thick, Merck), with UV detection at 254 and 300 nm . Medium-pressure column chromatography was undertaken on a $530-4-10 \mathrm{~V}$ apparatus (Yamazen) with Lobar grosse B (310-25, Lichroprep Si60, Merck) as column absorbent. For flash column chromatography, Merck Kieselgel 60 ( $230-400$ mesh) was used. Ether refers to diethyl ether. Light petroleum refers to the fraction boiling in the range $80-110^{\circ} \mathrm{C}$.
$(1 \alpha, 2 \alpha, 4 a \alpha, 10 b \alpha)-( \pm)-1,2,3,4,4 a, 5,6,10 b-O c t a h y d r o-1-h y d r-$ oxy-4-methylbenzo[f] quinolin-2-ylmethanol 8.-Ozone gas was slowly bubbled into a solution of the cis-lactam $3^{7 a}(300 \mathrm{mg})$ in methanol ( $20 \mathrm{~cm}^{3}$ ) at $-30^{\circ} \mathrm{C}$ in the presence of Oil Violet ${ }^{15}$ $(1-2 \mathrm{mg})$ until the violet colour disappeared ( 5 min ). Removal of the solvent gave a residue, which was dissolved in anhydrous THF ( $50 \mathrm{~cm}^{3}$ ), and this solution was added dropwise to a suspension of lithium aluminium hydride ( 500 mg ) in anhydrous ether ( $50 \mathrm{~cm}^{3}$ ) under reflux. The mixture was refluxed for an additional 2 h , and then treatment in the usual way ${ }^{9 a}$ gave a crystalline residue, which was recrystallised from methanol to give the diol $8(220 \mathrm{mg}, 76 \%$ ) as crystals, m.p. 189$190^{\circ} \mathrm{C} ; \delta_{\mathbf{H}}(200 \mathrm{MHz}) 7.33(1 \mathrm{H}, \mathrm{m}, 10-\mathrm{H}), 7.28-7.12(3 \mathrm{H}$, $\mathrm{m}, 7-, 8-\mathrm{and} 9-\mathrm{H}), 4.73(1 \mathrm{H}, \mathrm{br} \mathrm{t}, J 4,1-\mathrm{H}), 3.91(2 \mathrm{H}, \mathrm{s}$, $\mathrm{OH} \times 2), 3.73\left(2 \mathrm{H}, \mathrm{d}, J 5, \mathrm{CH}_{2} \mathrm{OH}\right), 3.08(1 \mathrm{H}$, br t,$J 4$, $10 \mathrm{~b}-\mathrm{H}), 2.96\left(1 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}^{\mathrm{eq}}\right), 2.90(1 \mathrm{H}, \mathrm{m}, 4 \mathrm{a}-\mathrm{H}), 2.70(1 \mathrm{H}, \mathrm{br}$ ddd, $J 17,6$ and $\left.3 \mathrm{~Hz}, 6-\mathrm{H}^{\mathrm{ax}}\right), 2.60-2.48\left(2 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}_{2}\right), 2.34$ ( $3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}$ ), $2.26\left(1 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}^{\mathrm{eq}}\right.$ ) and $1.90-1.64(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}$ and $5-\mathrm{H}^{\mathrm{ax}}$ ) (Found: $\mathrm{C}, 72.7 ; \mathrm{H}, 8.9 ; \mathrm{N}, 5.5 . \mathrm{C}_{15} \mathrm{H}_{21} \mathrm{NO}_{2}$ requires C, $72.8 ; \mathrm{H}, 8.6 ; \mathrm{N}, 5.7 \%$ ).
$(1 x, 2 x, 4 a x, 10 b x)-( \pm)-1,2,3,4,4 a, 5,6,10 b-O c t a h y d r o-1-h y d r o x y-$ 4-methylbenzo[f]quinolin-2-ylmethyl Acetate 9.-Acetic anhydride $\left(0.03 \mathrm{~cm}^{3}\right)$ was added dropwise to a stirred solution of the diol $8(100 \mathrm{mg})$ and DMAP ( 52 mg ) in ice-cooled methylene dichloride ( $20 \mathrm{~cm}^{3}$ ), and the mixture was stirred at $0^{\circ} \mathrm{C}$ for an additional 3.5 h . Then further methylene dichloride was added and the solution was washed successively with $10 \%$ aq. sodium carbonate and water. The organic layer was dried and evaporated to give a residue, which was purified by PLC (ethyl acetate) to afford the acetate 9 ( $87 \mathrm{mg}, 74 \%$ ) as an amorphous solid, $v_{\text {max }} / \mathrm{cm}^{-1} 3512(\mathrm{OH})$ and $1730(\mathrm{OAc}) ; \delta_{\mathrm{H}}(200 \mathrm{MHz})$ $7.32-7.12(4 \mathrm{H}, \mathrm{m}, \mathrm{ArH}), 4.55(1 \mathrm{H}, \mathrm{br} \mathrm{t}, J 3,1-\mathrm{H}), 4.38(1 \mathrm{H}, \mathrm{dd}$, $J 11$ and $7, \mathrm{CHHOAc}), 3.95(1 \mathrm{H}$, dd, $J 11$ and $7, \mathrm{CHHOAc}$ ), $3.08(1 \mathrm{H}$, br dd, $J 4$ and $3,10 \mathrm{~b}-\mathrm{H}), 3.00\left(1 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}^{\mathrm{eq}}\right), 2.83$ $(1 \mathrm{H}, \mathrm{m}, 4 \mathrm{a}-\mathrm{H}), 2.69\left(1 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}^{\mathrm{ax}}\right), 2.58(1 \mathrm{H}, \mathrm{dd}, J 11$ and 4 , $\left.3-\mathrm{H}^{\mathrm{eq}}\right), 2.42\left(1 \mathrm{H}, \mathrm{t}, J 11,3-\mathrm{H}^{\mathrm{ax}}\right), 2.36(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}), 2.25(1 \mathrm{H}$, $\left.\mathrm{m}, 5-\mathrm{H}^{\mathrm{eq}}\right), 2.11(3 \mathrm{H}, \mathrm{s}, \mathrm{OAc}), 2.04(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H})$ and $1.71(1 \mathrm{H}$, $\mathrm{m}, 5-\mathrm{H}^{\mathrm{ax}}$ ) (Found: $\mathrm{M}^{+}$, 289.167. $\mathrm{C}_{17} \mathrm{H}_{23} \mathrm{NO}_{3}$ requires M , 289.168).
$(1 \alpha, 2 \alpha, 4 a \alpha, 10 b \alpha)-( \pm)-1,2,3,4,4 a, 5,6,10 b-O c t a h y d r o-1-h y d r-$ oxy-4-methylbenzo[f]quinolin-2-ylmethyl Methanesulphonate 10.-Mesyl chloride $\left(0.1 \mathrm{~cm}^{3}\right)$ was added dropwise to a stirred solution of the diol $8(250 \mathrm{mg})$ in ice-cooled pyridine ( $3 \mathrm{~cm}^{3}$ ), and the mixture was stirred at $0^{\circ} \mathrm{C}$ for an additional 1 h . Then $10 \%$ aq. ammonium hydroxide was added, and the mixture was extracted with methylene dichloride. The extract was washed, dried and evaporated. The residue was purified by flash chromatography (methylene dichloride) to give the monomesyl derivative $10\left(190 \mathrm{mg}, 80 \%\right.$ ) as crystals, m.p. $120-122^{\circ} \mathrm{C}$ (from methylene dichloride-ether); $v_{\text {max }} / \mathrm{cm}^{-1} 3608(\mathrm{OH}) ; 1360$ and $1174\left(\mathrm{SO}_{2}\right) ; \delta_{\mathrm{H}}(60 \mathrm{MHz}) 4.50(1 \mathrm{H}$, br dd, $J 4$ and $2,1-\mathrm{H}), 2.97$ ( $3 \mathrm{H}, \mathrm{s}, \mathrm{Ms}$ ) and 2.31 ( $3 \mathrm{H}, \mathrm{s}$, NMe) (Found: C, $58.4 ; \mathrm{H}, 7.0 ; \mathrm{N}$, 4.4. $\mathrm{C}_{16} \mathrm{H}_{23} \mathrm{NO}_{4} \mathrm{~S} \cdot \frac{1}{2} \overline{0} \mathrm{CH}_{2} \mathrm{Cl}_{2}$ requires $\mathrm{C}, 58.5 ; \mathrm{H}, 7.1 ; \mathrm{N}, 4.2 \%$ ).
$(1 \alpha, 2 \alpha, 4 a \alpha, 10 b \alpha)-( \pm)-1,2,3,4,4 a, 5,6,10 b-O c t a h y d r o-2,4-d i-$ methylbenzo[f]quinolin-1-ol 11.-Sodium borohydride (130 mg ) was added to a stirred solution of the monomesyl compound $10(190 \mathrm{mg})$ in DMSO $\left(3 \mathrm{~cm}^{3}\right)$ at $10^{\circ} \mathrm{C}$, and the mixture was heated at $80^{\circ} \mathrm{C}$ for 8 h . After being cooled the reaction mixture was treated with water, and this mixture was extracted with ethyl acetate. The extract was washed, dried and evaporated. The residue was purified by PLC (methylene dichloride-methanol, 9:1) to afford the alcohol 11 ( $84 \mathrm{mg}, 62 \%$ ) as crystals, m.p. $100-101^{\circ} \mathrm{C}$ (from light petroleum); $v_{\text {max }} / \mathrm{cm}^{-1}$ $3616(\mathrm{OH}) ; \delta_{\mathrm{H}}(200 \mathrm{MHz}) 7.34-7.10(4 \mathrm{H}, \mathrm{m}, \operatorname{ArH}), 4.45(1 \mathrm{H}$, br d, $J 3,1-\mathrm{H}), 3.08(1 \mathrm{H}$, br s, $10 \mathrm{~b}-\mathrm{H}), 3.00\left(1 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}^{\text {eq }}\right)$,
$2.75(1 \mathrm{H}$, br s, $4 \mathrm{a}-\mathrm{H}), 2.64\left(1 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}^{\mathrm{ax}}\right), 2.53(1 \mathrm{H}, \mathrm{dd}, J 12$ and 4, $3-\mathrm{H}^{\mathrm{eq}}$ ), $2.30(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}), 2.26\left(1 \mathrm{H}, \mathrm{t}, J 12,3-\mathrm{H}^{\mathrm{ax}}\right), 2.24$ $\left(1 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}^{\mathrm{eq}}\right), 1.94-1.60\left(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}\right.$ and $\left.5-\mathrm{H}^{\mathrm{ax}}\right)$ and 0.97 ( $3 \mathrm{H}, \mathrm{d}, J 7,2-\mathrm{Me}$ ) (Found: C, 77.7; H, 9.2; N, 6.0. $\mathrm{C}_{15} \mathrm{H}_{21} \mathrm{NO}$ requires $\mathrm{C}, 77.9 ; \mathrm{H}, 9.15 ; \mathrm{N}, 6.05 \%$ ).

Methylation of the Lactam 2.-(a) A solution of LDA, prepared from diisopropylamine $\left(0.28 \mathrm{~cm}^{3}\right)$ and butyllithium ( $1.57 \mathrm{~mol} \mathrm{dm}^{-3}$ solution in hexane) $\left(1.2 \mathrm{~cm}^{3}\right)$ at $-40^{\circ} \mathrm{C}$, was added dropwise to a stirred solution of the lactam $2(100 \mathrm{mg})$ and methyl iodide ( 0.073 ml ) in anhydrous THF ( $10 \mathrm{~cm}^{3}$ ) at $-40^{\circ} \mathrm{C}$. Addition of LDA was stopped when the red colour of the reaction mixture became permanent. After being stirred at $-40^{\circ} \mathrm{C}$ for 10 min the mixture was treated with water and was then extracted with methylene dichloride. The extract was washed, dried and evaporated to give a residue, which was chromatographed on a medium-pressure column (ethyl acetatehexane, $3: 2$ ) to afford ( $3 \mathrm{a} x, 5 \mathrm{a} \alpha, 11 \mathrm{~b} \beta, 11 \mathrm{c} x$ )-( $\pm$ )-5,5a,6,7,11b,11c-hexahydro-3a,5-dimethylbenzo[f] furo $[3,2-\mathrm{c}]$ quinolin- $4(3 a \mathrm{H})$ one 12 ( $90 \mathrm{mg}, 87 \%$ ) as crystals, m.p. $118-120^{\circ} \mathrm{C}$ (from ether); $v_{\text {max }} / \mathrm{cm}^{-1} 1640(\mathrm{NCO}) ; \delta_{\mathrm{H}}(200 \mathrm{MHz}) 7.90(1 \mathrm{H}, \mathrm{m}, 11-\mathrm{H}:$ $24 \%$ intensity increase upon irradiation at $\delta 1.44$ ), 7.34-7.08 ( $3 \mathrm{H}, \mathrm{m}, 8-, 9-\mathrm{and} 10-\mathrm{H}), 6.40(1 \mathrm{H}, \mathrm{d}, J 3,2-\mathrm{H}), 5.37(1 \mathrm{H}, \mathrm{d}$, $J 3,3-\mathrm{H}), 4.46(1 \mathrm{H}, \mathrm{d}, J 10,11 \mathrm{c}-\mathrm{H}), 3.48(1 \mathrm{H}$, ddd, $J 13,11$ and $3,5 \mathrm{a}-\mathrm{H}), 3.18-2.82\left(3 \mathrm{H}, \mathrm{m}, 7-\mathrm{H}_{2}\right.$ and $\left.11 \mathrm{~b}-\mathrm{H}\right), 3.11(3 \mathrm{H}, \mathrm{s}$, NMe), $2.56\left(1 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}^{\mathrm{eq}}\right), 1.86\left(1 \mathrm{H}\right.$, br qd, $J 12$ and $\left.6.5,6-\mathrm{H}^{\mathrm{ax}}\right)$ and $1.44(3 \mathrm{H}, \mathrm{s}, 3 \mathrm{a}-\mathrm{Me}: 2 \%$ intensity increase upon irradiation at $\delta 4.46$ ) (Found: $\mathrm{C}, 75.7 ; \mathrm{H}, 7.2 ; \mathrm{N}, 5.1 . \mathrm{C}_{17} \mathrm{H}_{19} \mathrm{NO}_{2}$ requires C, 75.8; H, 7.1; N, 5.2\%).
(b) A solution of the lactam $2(550 \mathrm{mg})$ in anhydrous THF ( $10 \mathrm{~cm}^{3}$ ) was added at $-78^{\circ} \mathrm{C}$ to a stirred solution of LDA, prepared from diisopropylamine ( $0.25 \mathrm{~cm}^{3}$ ) and butyllithium $\left(1.55 \%\right.$ mol dm${ }^{-3}$ solution in hexane) $\left(1.3 \mathrm{~cm}^{3}\right)$ at $-78^{\circ} \mathrm{C}$, and the mixture was stirred at $-78^{\circ} \mathrm{C}$ for 10 min , then methyl iodide ( $1 \mathrm{~cm}^{3}$ ) was added. After the mixture had been stirred at $-78^{\circ} \mathrm{C}$ for 10 min and then at $-40^{\circ} \mathrm{C}$ for 40 min , water was added and the reaction mixture was extracted with methylene dichloride. The extract was washed, dried and evaporated. The residue was purified as above $(a)$ to give the same methyl lactam 12 ( $314 \mathrm{mg}, 54 \%$ ) and 2,3,4,4a,5,6-hexahydro-2,4-dimethyl-3-oxobenzo[f]quinolin-2-ylacetaldehyde $13(11 \mathrm{mg}$, $2 \%$ ) as a pale yellow oil, $v_{\max } / \mathrm{cm}^{-1} 1725(\mathrm{CHO})$ and 1630 $(\mathrm{NCO}) ; \delta_{\mathrm{H}}(200 \mathrm{MHz}) 9.75(1 \mathrm{H}, \mathrm{dd}, J 3$ and $1, \mathrm{CHO}), 7.48$ ( $1 \mathrm{H}, \mathrm{m}, 10-\mathrm{H}), 7.32-7.12(3 \mathrm{H}, \mathrm{m}, 7-, 8-\mathrm{and} 9-\mathrm{H}), 5.86(1 \mathrm{H}, \mathrm{d}$, $J 1.5,1-\mathrm{H}), 4.10(1 \mathrm{H}$, br ddd, $J 13,3$ and $1.5,4 \mathrm{a}-\mathrm{H}), 3.16-3.00$ $\left(2 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}_{2}\right), 3.12(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}), 3.11(1 \mathrm{H}, \mathrm{dd}, J 16.5$ and 1 , $\mathrm{CHHCHO}), 2.61(1 \mathrm{H}$, dd,$J 16.5$ and $3, \mathrm{CH} H \mathrm{CHO}), 2.60(1 \mathrm{H}$, $\left.\mathrm{m}, 5-\mathrm{H}^{\mathrm{eq}}\right), 1.80\left(1 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}^{\mathrm{ax}}\right)$ and $1.45(3 \mathrm{H}, \mathrm{s}, 2-\mathrm{Me})$ (Found: $\mathrm{M}^{+}, 269.141 . \mathrm{C}_{17} \mathrm{H}_{19} \mathrm{NO}_{2}$ requires $\mathrm{M}, 269.141$ ).
$(1 \alpha, 2 \alpha, 4 a \beta, 10 b \alpha)-( \pm)-1,2,3,4,4 a, 5,6,10 b-O c t a h y d r o-1-h y d r-$ oxy-2,4-dimethylbenzo[f]quinolin-2-ylmethanol 14.-According to the procedure given for the preparation of compound $\mathbf{8}$, ozonolysis of the lactam $12(150 \mathrm{mg})$ in methanol $\left(10 \mathrm{~cm}^{3}\right)$ in the presence of Oil Violet $(1-2 \mathrm{mg})$ was followed by reduction with lithium aluminium hydride ( 350 mg ) in anhydrous etherTHF ( $1: 1 ; 50 \mathrm{~cm}^{3}$ ). The crude product was purified by flash chromatography (methylene dichloride-methanol, 9:1) to afford the diol $14\left(119 \mathrm{mg}, 79 \%\right.$ ) as crystals, m.p. $128-132.5^{\circ} \mathrm{C}$ (from ether); $\delta_{\mathrm{H}}(200 \mathrm{MHz}) 7.77(1 \mathrm{H}, \mathrm{m}, 10-\mathrm{H}), 7.32-7.06$ ( $3 \mathrm{H}, \mathrm{m}, 7-, 8-$ and $9-\mathrm{H}$ ), $3.96(1 \mathrm{H}, \mathrm{dd}, J 11$ and $2.5, \mathrm{CHHOH}$ ), $3.81(1 \mathrm{H}, \mathrm{d}, J 10.5,1-\mathrm{H}), 3.73(1 \mathrm{H}, \mathrm{d}, J 11, \mathrm{CH} H \mathrm{OH}), 3.11$ ( $1 \mathrm{H}, \mathrm{t}, J 10.5,10 \mathrm{~b}-\mathrm{H}), 2.87-2.66\left(2 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}_{2}\right), 2.86(1 \mathrm{H}, \mathrm{d}$, $\left.J 12,3-\mathrm{H}^{\mathrm{eq}}\right), 2.22\left(1 \mathrm{H}\right.$, dd, $J 12$ and $\left.2.5,3-\mathrm{H}^{\mathrm{ax}}\right), 2.20(3 \mathrm{H}, \mathrm{s}$, NMe), $2.05\left(1 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}^{\mathrm{eq}}\right), 1.85-1.53\left(2 \mathrm{H}, \mathrm{m}, 4 \mathrm{a}-\mathrm{H}\right.$ and $\left.5-\mathrm{H}^{\mathrm{ax}}\right)$ and $0.92(3 \mathrm{H}, \mathrm{s}, 2-\mathrm{Me})$ (Found: $\mathrm{C}, 73.5 ; \mathrm{H}, 9.1 ; \mathrm{N}, 5.4$. $\mathrm{C}_{16} \mathrm{H}_{23} \mathrm{NO}_{2}$ requires $\mathrm{C}, 73.5 ; \mathrm{H}, 8.9 ; \mathrm{N}, 5.4 \%$ ).
$(1 \alpha, 2 \alpha, 4 a \beta, 10 b \alpha)-( \pm)-1,2,3,4,4 a, 5,6,10 b-O c t a h y d r o-1-h y d r-$ oxy-2,4-dimethylbenzo[f]quinolin-2-ylmethyl Acetate 15.-According to the acetylation procedure described for compound 8 , treatment of the diol $\mathbf{1 4}(98 \mathrm{mg})$ in methylene dichloride ( 17 $\mathrm{cm}^{3}$ ) with acetic anhydride ( $0.04 \mathrm{~cm}^{3}$ ) in the presence of DMAP $(52.4 \mathrm{mg})$, followed by recrystallisation of the crude product from ethyl acetate, gave the acetate $15(113 \mathrm{mg}, 99 \%$ ) as crystals, m.p. $148.5-149.5^{\circ} \mathrm{C} ; v_{\max } / \mathrm{cm}^{-1} 1735(\mathrm{OAc}) ; \delta_{\mathrm{H}}(200 \mathrm{MHz})$ $7.64(1 \mathrm{H}, \mathrm{m}, 10-\mathrm{H}), 7.32-7.10(3 \mathrm{H}, \mathrm{m}, 7-, 8-\mathrm{and} 9-\mathrm{H}), 4.64$ and $4.44\left(2 \mathrm{H}, \mathrm{ABq}, J 11, \mathrm{CH}_{2} \mathrm{OAc}\right), 3.84(1 \mathrm{H}$, br d$, J 11,1-\mathrm{H}), 2.98-$ $2.68\left(2 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}_{2}\right), 2.76\left(1 \mathrm{H}, \mathrm{d}, J 12,3-\mathrm{H}^{\mathrm{eq}}\right), 2.71(1 \mathrm{H}, \mathrm{t}, J 11$, $10 \mathrm{~b}-\mathrm{H}), 2.18(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}), 2.10(3 \mathrm{H}, \mathrm{s}, \mathrm{OAc}), 2.02(1 \mathrm{H}, \mathrm{m}$, $\left.5-\mathrm{H}^{\mathrm{eq}}\right), 1.90\left(1 \mathrm{H}, \mathrm{d}, J 12,3-\mathrm{H}^{\mathrm{ax}}\right), 1.90-1.58(2 \mathrm{H}, \mathrm{m}, 4 \mathrm{a}-\mathrm{H}$ and $5-\mathrm{H}^{\mathrm{ax}}$ ), 1.16 ( $3 \mathrm{H}, \mathrm{s}, 2-\mathrm{Me}$ ) (Found: C, $71.1 ; \mathrm{H}, 8.45$; N, 4.5. $\mathrm{C}_{18} \mathrm{H}_{25} \mathrm{NO}_{3}$ requires $\mathrm{C}, 71.25 ; \mathrm{H}, 8.3 ; \mathrm{N}, 4.6 \%$ ).
$(1 \alpha, 2 \alpha, 4 a \alpha, 10 b \alpha)-( \pm)-1,2,3,4,4 a, 5,6,10 b-O c t a h y d r o-4-m e t h y l-$ 1-methylsulfonyloxybenzo[f]quinolin-2-ylmethyl Acetate 18.Mesyl chloride ( $0.2 \mathrm{~cm}^{3}$ ) was added dropwise to a stirred, icecooled solution of the acetate $9(40 \mathrm{mg})$ in pyridine $\left(1 \mathrm{~cm}^{3}\right)$, and the mixture was then stirred at room temperature for an additional 2 h . Then $10 \%$ aq. ammonium hydroxide was added, and the mixture was extracted with methylene dichloride. The extract was washed, dried and evaporated to give a crystalline residue, which was recrystallised from ether to afford the monomesyl derivative 18 ( $40 \mathrm{mg}, 79 \%$ ) as crystals, m.p. $100-$ $101{ }^{\circ} \mathrm{C} ; \delta_{\mathbf{H}}(200 \mathrm{MHz}) 7.37-7.12(4 \mathrm{H}, \mathrm{m}, \mathrm{ArH}), 5.61(1 \mathrm{H}, \mathrm{br} \mathrm{s}$, $1-\mathrm{H}), 4.19(1 \mathrm{H}, \mathrm{dd}, J 11$ and $8, \mathrm{CH} \mathrm{HOAc}), 4.04(1 \mathrm{H}, \mathrm{dd}, J 11$ and $7, \mathrm{CH} H \mathrm{OAc}), 3.46(1 \mathrm{H}, \mathrm{br}$ s, 10b-H), $3.06(3 \mathrm{H}, \mathrm{s}, \mathrm{Ms}), 3.02$ $\left(1 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}^{\mathrm{eq}}\right), 2.87(1 \mathrm{H}, \mathrm{br} s, 4 \mathrm{a}-\mathrm{H}), 2.75-2.58\left(2 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}^{\mathrm{ax}}\right.$ and $\left.3-\mathrm{H}^{\mathrm{eq}}\right), 2.45-2.16\left(3 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}, 3-\mathrm{H}^{\mathrm{ax}}\right.$ and $\left.5-\mathrm{H}^{\mathrm{eq}}\right), 2.38(3 \mathrm{H}$, $\mathrm{s}, \mathrm{NMe}), 2.10(3 \mathrm{H}, \mathrm{s}, \mathrm{Ac})$ and $1.75\left(1 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}^{\mathrm{ax}}\right)$ (Found: C, $59.1 ; \mathrm{H}, 7.0 ; \mathrm{N}, 3.8 . \mathrm{C}_{18} \mathrm{H}_{25} \mathrm{NO}_{5} \mathrm{~S}$ requires $\mathrm{C}, 58.8 ; \mathrm{H}, 6.9 ; \mathrm{N}$, $3.8 \%$ ).
$(1 x, 2 x, 4 a \alpha, 10 b \alpha)-( \pm)-1,2,3,4,4 a, 5,6,10 b-O c t a h y d r o-2,4-d i-$ methylbenzo[f]quinolin-1-yl Methanesulfonate 19.-According to the mesylation procedure described for compound 9 , treatment of the alcohol $11(145 \mathrm{mg})$ in pyridine $\left(3 \mathrm{~cm}^{3}\right)$ with mesyl chloride ( $0.1 \mathrm{~cm}^{3}$ ), followed by purification by PLC (methylene dichloride-methanol, 98:2), gave the monomesyl derivative 19 ( $150 \mathrm{mg}, 78 \%$ ) as crystals, m.p. $95-97^{\circ} \mathrm{C}$ (from ether); $v_{\text {max }} / \mathrm{cm}^{-1} 1352,1334$ and $1174(\mathrm{OMs}) ; \delta_{\mathrm{H}}(60 \mathrm{MHz}) 5.33$ ( $1 \mathrm{H}, \mathrm{m}, 1-\mathrm{H}$ ), $3.40(1 \mathrm{H}, \mathrm{br} \mathrm{t}, J 3,10 \mathrm{~b}-\mathrm{H}), 3.00(3 \mathrm{H}, \mathrm{s}, \mathrm{Ms}), 2.30$ ( $3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}$ ) and $1.03(3 \mathrm{H}, \mathrm{d}, J 6,2-\mathrm{Me})$ (Found: C, $61.9 ; \mathrm{H}$, 7.4; $\mathrm{N}, 4.3 . \mathrm{C}_{16} \mathrm{H}_{23} \mathrm{NO}_{3} \mathrm{~S} \mathrm{C}, 62.1 ; \mathrm{H}, 7.5 ; \mathrm{N}, 4.5 \%$ ).
$(1 x, 2 x, 4 a \beta, 10 b \alpha)-( \pm)-1,2,3,4,4 a, 5,6,10 b-O c t a h y d r o-2,4-d i-$ methyl-1-methylsulfonyloxybenzo[f]quinolin-2-ylmethyl Acetate 20.-Mesyl chloride ( 114.6 mg ) was added dropwise to a stirred, ice-cooled solution of the acetate $15(202 \mathrm{mg})$ in toluene ( $40 \mathrm{~cm}^{3}$ ) in the presence of triethylamine $(0.17 \mathrm{ml})$, and the mixture was then stirred at $0^{\circ} \mathrm{C}$ for 1 h . Then $10 \%$ aq. ammonium hydroxide was added, and the mixture was extracted with benzene. The extract was washed, dried and evaporated to give a crystalline residue, which was recrystallised from ether-methanol to afford the mesate $20(222 \mathrm{mg}, 87 \%)$ as crystals, m.p. $189-192^{\circ} \mathrm{C} ; \delta_{\mathrm{H}}(200 \mathrm{MHz}) 7.48(1 \mathrm{H}, \mathrm{m}, 10-\mathrm{H})$, $7.38-7.15(3 \mathrm{H}, \mathrm{m}, 7-, 8-$ and $9-\mathrm{H}), 5.19(1 \mathrm{H}, \mathrm{d}, J 11,1-\mathrm{H}), 4.56$ and $4.41\left(2 \mathrm{H}, \mathrm{ABq}, J 11, \mathrm{CH}_{2} \mathrm{OAc}\right), 3.10-2.65\left(4 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}^{\mathrm{eq}}\right.$, $6-\mathrm{H}_{2}$ and $\left.10 \mathrm{~b}-\mathrm{H}\right), 2.98(3 \mathrm{H}, \mathrm{s}, \mathrm{Ms}), 2.17(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}), 2.12$ $(3 \mathrm{H}, \mathrm{s}, \mathrm{OAc}), 2.06-1.56\left(4 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}^{\mathrm{ax}}, 4 \mathrm{a}-\mathrm{H}\right.$ and $\left.5-\mathrm{H}_{2}\right)$ and $1.32(3 \mathrm{H}, \mathrm{s}, 2-\mathrm{Me})$ (Found: $\mathrm{M}^{+}, 381.160 . \mathrm{C}_{19} \mathrm{H}_{27} \mathrm{NO}_{5} \mathrm{~S}$ requires M, 381.161).

Fragmentation of the Acetate 4.-(a) Mesyl chloride ( 0.1 $\mathrm{cm}^{3}$ ) was added dropwise to a stirred solution of the acetate $4(29 \mathrm{mg})$ in pyridine $\left(1.5 \mathrm{~cm}^{3}\right)$ at $0^{\circ} \mathrm{C}$, and the mixture was stirred at $0^{\circ} \mathrm{C}$ for an additional 3.5 h . Then $10 \%$ aq. ammonium
hydroxide was added and the mixture was extracted with methylene dichloride. The extract was washed, dried and evaporated to give a residue, which was separated by PLC (methylene dichloride-methanol, 19:1) to afford ( $1 \alpha, 2 \beta, 4 a \beta$, $10 b \alpha)$-( $\pm$ )-1-chloro-1,2,3,4,4a,5,6,10b-octahydro-4-methylbenzo[ $f$ ]quinolin-2-ylmethyl acetate $21(1 \mathrm{mg}, 5 \%)$ and $\left[1^{\prime} \alpha(\mathrm{E}), 2^{\prime} \beta\right]-( \pm)-3-\left[1^{\prime}, 2^{\prime}, 3^{\prime}, 4^{\prime}\right.$-tetrahydro- $2^{\prime}$-(methylamino) $-1^{\prime}$ naphthyl] prop-2-enyl acetate $22(9 \mathrm{mg}, 34 \%)$ as an oil, $v_{\text {max }} / \mathrm{cm}^{-1} 3320(\mathrm{NH})$ and $1732(\mathrm{OAc}) ; \delta_{\mathrm{H}}(200 \mathrm{MHz}) 7.25-7.10$ $(4 \mathrm{H}, \mathrm{m}, \mathrm{ArH}), 5.78(1 \mathrm{H}, \mathrm{dt}, J 15$ and $5.5,2-\mathrm{H}), 5.71(1 \mathrm{H}$, dd, $J 15$ and $8,3-\mathrm{H}), 4.67\left(2 \mathrm{H}\right.$, br d, $\left.J 5, \mathrm{CH}_{2} \mathrm{OAc}\right), 3.50(1 \mathrm{H}$, br t, $\left.J 8,1^{\prime}-\mathrm{H}\right), 3.00-2.82\left(2 \mathrm{H}, \mathrm{m}, 4^{\prime}-\mathrm{H}_{2}\right), 2.75(1 \mathrm{H}, \mathrm{ddd}, J 10,8$ and 3 , $\left.2^{\prime}-\mathrm{H}\right), 2.57(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}), 2.24\left(1 \mathrm{H}, \mathrm{m}, 3^{\prime}-\mathrm{H}^{\mathrm{eq}}\right), 2.10(3 \mathrm{H}, \mathrm{s}, \mathrm{OAc})$ and $1.72\left(1 \mathrm{H}, \mathrm{m}, 3^{\prime}-\mathrm{H}^{\text {ax }}\right)$ (Found: $\mathrm{M}^{+}$, 259.154. $\mathrm{C}_{16} \mathrm{H}_{21} \mathrm{NO}_{2}$ requires $\mathrm{M}, 259.157$ ). The chloride 21 was identical with the sample reported before ${ }^{7 b}$ upon comparison of their $R_{f}$-values and IR and ${ }^{1} \mathrm{H}$ NMR spectra.
(b) Mesylation of the acetate $4(29 \mathrm{mg})$ with mesyl chloride $\left(0.02 \mathrm{~cm}^{3}\right)$ in the presence of triethylamine $\left(0.05 \mathrm{~cm}^{3}\right)$ in either methylene dichloride $\left(3 \mathrm{~cm}^{3}\right)$ or toluene $\left(3 \mathrm{~cm}^{3}\right)$, and then purification by the same procedure ( $a$ ), gave the same chloride 21 and the same allyl acetate 22, as shown in Table 1.

Fragmentation of the Amine 5.-The amine $5(50 \mathrm{mg})$ was treated with mesyl chloride $\left(0.05 \mathrm{~cm}^{3}\right)$ in the presence of triethylamine $\left(0.1 \mathrm{~cm}^{3}\right)$ in toluene $\left(8 \mathrm{~cm}^{3}\right)$ as described above. The crude product was purified by PLC (ether-methanol, 1:1) to afford $[1 \alpha(\mathrm{E}), 2 \beta]-( \pm)-1,2,3,4-$ tetrahydro- N -methyl-1-(prop-$1^{\prime}$-enyl)naphthalen-2-amine $23(19 \mathrm{mg}, 44 \%)$ as an oil, $v_{\max } / \mathrm{cm}^{-1} 3316(\mathrm{NH}) ; \delta_{\mathrm{H}}(200 \mathrm{MHz}) 7.24-7.04(4 \mathrm{H}, \mathrm{m} \mathrm{ArH})$, $5.70\left(1 \mathrm{H}, \mathrm{dq}, J 15\right.$ and $\left.6,2^{\prime}-\mathrm{H}\right), 5.34\left(1 \mathrm{H}\right.$, ddq, $J 15,9$ and $2,1^{\prime}-$ H), $3.31(1 \mathrm{H}$, br t, $J 9,1-\mathrm{H}), 2.94-2.72\left(2 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}_{2}\right), 2.60(1 \mathrm{H}$, ddd, $J 10,8$ and $3,2-\mathrm{H}), 2.52(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}), 2.24\left(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}^{\mathrm{eq}}\right)$, $1.80\left(3 \mathrm{H}\right.$, dd, $J 6$ and $2, \mathrm{Me}$ ) and $1.64\left(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}^{\mathrm{ax}}\right)$ (Found: $\mathrm{M}^{+}, 201.146 . \mathrm{C}_{14} \mathrm{H}_{19} \mathrm{~N}$ requires $\mathrm{M}, 201.152$ ).

Fragmentation of the Mesyl Derivative 16.-The mesyl compound $16(50 \mathrm{mg})$ was treated in either $80 \%$ aq. ethanol ( $6 \mathrm{~cm}^{3}$ ) or absolute ethanol ( $6 \mathrm{~cm}^{3}$ ) or neat at the temperature and for the reaction time described in Table 1. After the reaction mixture had been ice-cooled, $10 \%$ aq. ammonium hydroxide was added, and the mixture was extracted with methylene dichloride. The organic layer was washed with water, dried and evaporated to give a residue, which was separated by PLC (ether-methanol, 1:1) to give the allyl acetates 22,24 and 25 in the isolated yields shown in Table 1.
[1' $\left.x(\mathrm{Z}), 2^{\prime} \mathrm{B}\right]-( \pm)-3-\left[1^{\prime}, 2^{\prime}, 3^{\prime}, 4^{\prime}\right.$-Tetrahydro- $2^{\prime}$-(methylamino)-1'-naphthyl] prop-2-enyl acetate 24: an oil, $v_{\text {max }} / \mathrm{cm}^{-1} 3328$ $(\mathrm{NH})$ and $1732(\mathrm{OAc}) ; \delta_{\mathrm{H}}(200 \mathrm{MHz}) 7.24-7.06(4 \mathrm{H}, \mathrm{m}, \mathrm{ArH})$, $5.86(1 \mathrm{H}, \mathrm{dt}, J 11$ and $7,2-\mathrm{H}), 5.52(1 \mathrm{H}, \mathrm{br} \mathrm{t}, J 11,3-\mathrm{H}), 4.84$ ( 2 H , dd, $J 7$ and $1, \mathrm{CH}_{2} \mathrm{OAc}$ ), $3.65\left(1 \mathrm{H}\right.$, dd, $J 11$ and $9,1^{\prime}-\mathrm{H}$ ), $3.00-2.83\left(2 \mathrm{H}, \mathrm{m}, 4^{\prime}-\mathrm{H}_{2}\right), 2.61\left(1 \mathrm{H}\right.$, ddd, $J 10,9$ and $\left.3,2^{\prime}-\mathrm{H}\right)$, 2.55 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}$ ), $2.30\left(1 \mathrm{H}, \mathrm{m}, 3^{\prime}-\mathrm{H}^{\mathrm{eq}}\right)$, $2.12(3 \mathrm{H}, \mathrm{s}, \mathrm{OAc})$ and $1.68\left(1 \mathrm{H}, \mathrm{m}, 3^{\prime}-\mathrm{H}^{\mathrm{ax}}\right.$ ) (Found: $\mathrm{M}^{+}$, 259.155. $\mathrm{C}_{16} \mathrm{H}_{21} \mathrm{NO}_{2}$ requires $\mathrm{M}, 259.157$ ).
$\left[1^{\prime} x(\mathrm{E}), 2^{\prime} x\right]-( \pm)-3-\left[1^{\prime}, 2^{\prime}, 3^{\prime}, 4^{\prime}-T e t r a h y d r o-2^{\prime}-(\right.$ methylamino $)$ -1'-naphthyl]prop-2-enyl acetate 25: an oil, $v_{\max } / \mathrm{cm}^{-1} 3320$ $(\mathrm{NH})$ and $1734(\mathrm{OAc}) ; \delta_{\mathrm{H}}(200 \mathrm{MHz}) 7.25-7.08(4 \mathrm{H}, \mathrm{m}, \mathrm{ArH})$, $5.91(1 \mathrm{H}$, dd, $J 16$ and $8.5,3-\mathrm{H}), 5.64(1 \mathrm{H}, \mathrm{dt}, J 16$ and $6,2-\mathrm{H})$, 4.57 ( 2 H , dd, $J 6$ and $1, \mathrm{CH}_{2} \mathrm{OAc}$ ), $3.74(1 \mathrm{H}$, dd, $J 8.5$ and 6, $\left.1^{\prime}-\mathrm{H}\right), 3.04-2.76\left(3 \mathrm{H}, \mathrm{m}, 2^{\prime}-\mathrm{H}\right.$ and $\left.4^{\prime}-\mathrm{H}_{2}\right), 2.53(3 \mathrm{H}, \mathrm{s}$, NMe), $2.05(3 \mathrm{H}, \mathrm{s}, \mathrm{OAc}), 1.98\left(1 \mathrm{H}, \mathrm{m}, 3^{\prime}-\mathrm{H}^{\mathrm{eq}}\right)$ and $1.78\left(1 \mathrm{H}, \mathrm{m}, 3^{\prime}-\mathrm{H}^{\mathrm{ax}}\right)$ (Found: $\mathrm{M}^{+}, 259.156$ ).

Fragmentation of the Mesyl Derivative 17.-The mesyl compound $17(50 \mathrm{mg})$ was treated in either $80 \%$ aq. ethanol $\left(6 \mathrm{~cm}^{3}\right)$ or absolute ethanol $\left(6 \mathrm{~cm}^{3}\right)$ or neat at the temperature and for the reaction time described in Table 1. Similar work-up
as above gave the amines $\mathbf{2 3}, 26$ and 27 in the isolated yields shown in Table 1. [1x(Z),2 3$]-( \pm)-1,2,3,4-$ Tetrahydro-N-methyl-1-(prop-1-enyl)naphthalen-2-amine 26: an oil, $v_{\text {max }} / \mathrm{cm}^{-1} 3316$ $(\mathrm{NH}) ; \delta_{\mathrm{H}}(200 \mathrm{MHz}) 7.20-7.08(4 \mathrm{H}, \mathrm{m}, \mathrm{ArH}), 5.89(1 \mathrm{H}, \mathrm{dqd}, J$ $10,6.5$, and $\left.0.5,2^{\prime}-\mathrm{H}\right), 5.34\left(1 \mathrm{H}, \mathrm{tq}, J 10\right.$ and $\left.2,1^{\prime}-\mathrm{H}\right), 3.69(1 \mathrm{H}$, br t, $J 10,1-\mathrm{H}), 2.92-2.73\left(2 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}_{2}\right), 2.55(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}), 2.50$ $(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}), 2.25\left(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}^{\mathrm{eq}}\right), 1.82(3 \mathrm{H}, \mathrm{dd}, J 6.5$ and 2 , $\mathrm{Me})$ and $1.60\left(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}^{\mathrm{ax}}\right)$ (Found: $\mathrm{M}^{+}$, 201.148. $\mathrm{C}_{14} \mathrm{H}_{19} \mathrm{~N}$ requires $\mathrm{M}, 201.152$ ).
$[1 x(\mathrm{E}), 2 x]-( \pm)-1,2,3,4-T e t r a h y d r o-N-m e t h y l-1-(p r o p-1 '$-en-yl)naphthalen-2-amine 27: an oil, $\nu_{\text {max }} / \mathrm{cm}^{-1} 3315$ (NH); $\delta_{\mathrm{H}}(500 \mathrm{MHz}) 7.22-7.05(4 \mathrm{H}, \mathrm{m}, \mathrm{ArH}), 5.57(1 \mathrm{H}, \mathrm{dq}, J 15$ and $\left.6,2^{\prime}-\mathrm{H}\right), 5.52\left(1 \mathrm{H}, \mathrm{br}\right.$ dd, $J 15$ and $\left.8,1^{\prime}-\mathrm{H}\right), 3.70(1 \mathrm{H}, \mathrm{dd}, J 8$ and $5,1-\mathrm{H}), 3.00-2.80\left(3 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}\right.$ and $\left.4-\mathrm{H}_{2}\right), 2.54(3 \mathrm{H}, \mathrm{s}$, NMe), $2.00\left(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}^{\mathrm{eq}}\right), 1.79\left(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}^{\mathrm{ax}}\right)$ and $1.71(3 \mathrm{H}$, d, J6, Me) (Found: M ${ }^{+}$, 201.148).

Fragmentation of the Mesyl Derivatives 18 and 19.-The mesylester 18 ( 50 mg ) or 19 ( 50 mg ) was treated in $80 \% \mathrm{aq}$. ethanol ( $6 \mathrm{~cm}^{3}$ ) or neat at the temperature and for the time described in Table 1. Similar work-up as given for the fragmentation of compound 16 gave the amines 22 and 25 from the mesate 18, and the amines 23 and 27 from the mesate 19, respectively, in the isolated yields shown in Table 1. These products were identical with the samples prepared above upon comparison of their $R_{\mathrm{f}}$-values and IR and ${ }^{1} \mathrm{H}$ NMR spectra.
(E)-( $\pm$ )-1,2,3,4,5,6-Hexahydro-4-methyl-4-benzazecin-6-ylmethyl Acetate 28.-(a) From compound 16. A solution of the mesyl compound $16(20 \mathrm{mg})$ in $80 \%$ aq. ethanol $\left(3 \mathrm{~cm}^{3}\right)$ was stirred at room temperature in the presence of sodium borohydride ( 60 mg ) for 2 h . The solvent was removed and water was added to the residue. The mixture was extracted with methylene dichloride. The extract was washed, dried and evaporated to give the hydrolysed amino alcohol, which was dissolved in pyridine ( $1 \mathrm{~cm}^{3}$ ), and acetic anhydride ( $0.2 \mathrm{~cm}^{3}$ ) was added to the resulting ice-cooled solution. The mixture was then stirred at room temperature for 3 h , when $10 \%$ aq. sodium carbonate was added and the mixture was extracted with methylene dichloride. The extract was washed, dried and evaporated to give a residue, which was purified by PLC (methylene dichloride-methanol, 19:1) to afford the benzazecine $28(12 \mathrm{mg}, 79 \%)$ as an oil, $v_{\text {max }} / \mathrm{cm}^{-1} 1734(\mathrm{OAc})$; $\delta_{\mathrm{H}}(200$ MHz) $7.32-7.10(4 \mathrm{H}, \mathrm{m}, \mathrm{ArH}), 6.90(1 \mathrm{H}, \mathrm{d}, J 16,8-\mathrm{H}), 5.26$ ( 1 H , dd, $J 16$ and 9, 7-H), $4.16\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{OAc}\right), 3.16-1.38$ ( $9 \mathrm{H}, \mathrm{m}, 1-, 2-, 3-\mathrm{and} 5-\mathrm{H}_{2}$, and $\left.6-\mathrm{H}\right), 2.42(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe})$ and $2.09(3 \mathrm{H}, \mathrm{s}, \mathrm{OAc})$ (Found: $\mathrm{M}^{+}, 273.173 . \mathrm{C}_{17} \mathrm{H}_{23} \mathrm{NO}_{2}$ requires M, 273.173).
(b) From compound 18. A solution of the acetate $\mathbf{1 8}(20 \mathrm{mg})$ in $80 \%$ aq. ethanol ( $3 \mathrm{~cm}^{3}$ ) was stirred in the presence of sodium borohydride ( 60 mg ) at room temperature for 8 h . Similar acetylation of the crude product as described above gave the benzazecine 28 ( $8 \mathrm{mg}, 55 \%$ ), which was identical with the sample prepared before upon comparison of their $R_{\mathrm{f}}$-values and IR and ${ }^{1} \mathrm{H}$ NMR spectra.

Fragmentation of the Mesyl Derivative 20.-The amine 20 ( 50 mg ) was treated in either $80 \%$ aq. ethanol $\left(6 \mathrm{~cm}^{3}\right)$ or absolute ethanol ( $6 \mathrm{~cm}^{3}$ ) or neat at the temperature and for the time described in the relevant entry in Table 1. After similar work-up to that given for the fragmentation of compound 16, the crude product was separated by PLC (ethyl acetate-triethylamine, 92:8) to afford the allyl acetates 29 and $\mathbf{3 0}$ in the isolated yields shown in Table 1.
$\left[1^{\prime} x(Z), 2^{\prime} \beta\right]-( \pm)-2-M e t h y l-3-\left[1^{\prime}, 2^{\prime}, 3^{\prime}, 4^{\prime}\right.$-tetrahydro- $2^{\prime}$-(meth-ylamino)-1'-naphthyl]prop-2-enyl acetate 29: pale yellow crystals, m.p. $146-149{ }^{\circ} \mathrm{C}$ (from ether); $v_{\max } / \mathrm{cm}^{-1} 1730(\mathrm{OAc})$; $\delta_{\mathrm{H}}(200 \mathrm{MHz}) 7.28-7.00(4 \mathrm{H}, \mathrm{m}, \mathrm{ArH}), 5.34(1 \mathrm{H}, \mathrm{br} \mathrm{d}, J 10$,

3-H), 4.86 and $4.74\left(2 \mathrm{H}, \mathrm{ABq}, J 13, \mathrm{CH}_{2} \mathrm{OAc}\right), 3.77(1 \mathrm{H}, \mathrm{brt}$, $\left.J 10,1^{\prime}-\mathrm{H}\right), 3.00-2.82\left(2 \mathrm{H}, \mathrm{m}, 4^{\prime}-\mathrm{H}_{2}\right), 2.63(1 \mathrm{H}, \mathrm{ddd}, J 10,9$ and $\left.3,2^{\prime}-\mathrm{H}\right), 2.56(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}), 2.26\left(1 \mathrm{H}, \mathrm{m}, 3^{\prime}-\mathrm{H}^{\mathrm{eq}}\right), 2.11$ ( $3 \mathrm{H}, \mathrm{s}, \mathrm{OAc}$ ), $1.89(3 \mathrm{H}, \mathrm{d}, J 2, \mathrm{Me})$ and $1.70\left(1 \mathrm{H}, \mathrm{m}, 3^{\prime}-\mathrm{H}^{\mathrm{ax}}\right)$ (Found: $\mathrm{M}^{+}, 273.175 . \mathrm{C}_{17} \mathrm{H}_{23} \mathrm{NO}_{2}$ requires $\mathrm{M}, 273.173$ ).
$\left[1^{\prime} x(\mathrm{E}), 2^{\prime} \beta\right]-( \pm)-2-M e t h y l-3-\left[1^{\prime}, 2^{\prime}, 3^{\prime}, 4^{\prime}-\right.$ tetrahydro- $2^{\prime}$-(meth-ylamino)-1'-naphthyl]prop-2-enyl acetate 30: a pale yellow oil, $v_{\max } / \mathrm{cm}^{-1} 1730(\mathrm{OAc}) ; \delta_{\mathrm{H}}(200 \mathrm{MHz}) 7.24-7.00(4 \mathrm{H}, \mathrm{m}$, $\mathrm{ArH}), 5.41(1 \mathrm{H}, \mathrm{br} \mathrm{d}, J 10,3-\mathrm{H}), 4.62\left(2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{OAc}\right), 3.78$ $\left(1 \mathrm{H}\right.$, br $\left.\mathrm{t}, J 9.5,1^{\prime}-\mathrm{H}\right), 3.02-2.80\left(2 \mathrm{H}, \mathrm{m}, 4^{\prime}-\mathrm{H}_{2}\right), 2.70(1 \mathrm{H}$, br ddd, $J 10,9$ and $\left.3,2^{\prime}-\mathrm{H}\right), 2.54(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}), 2.28(1 \mathrm{H}, \mathrm{m}$, $\left.3^{\prime}-\mathrm{H}^{\mathrm{eq}}\right), 2.10(3 \mathrm{H}, \mathrm{s}, \mathrm{OAc}), 1.90(3 \mathrm{H}, \mathrm{d}, J 1.5, \mathrm{Me})$ and 1.75 $\left(1 \mathrm{H}, \mathrm{m}, 3^{\prime}-\mathrm{H}^{\mathrm{ax}}\right)$ (Found: $\mathrm{M}^{+}, 273.175$ ).
(E)-( $\pm$ )-1,2,3,4,5,6-Hexahydro-4,6-dimethyl-4-benzazecin-6y/methyl Acetate 31.-A solution of the mesyl compound 20 (9 $\mathrm{mg})$ in $80 \%$ aq. ethanol $\left(1 \mathrm{~cm}^{3}\right)$ was stirred in the presence of sodium borohydride ( 20 mg ) at room temperature for 1 h . The solvent was removed and water was added to the residue. The mixture was extracted with methylene dichloride. The extract was washed, dried and evaporated to give a residue, which was dissolved in pyridine ( $0.5 \mathrm{~cm}^{3}$ ), and acetic anhydride $\left(0.1 \mathrm{~cm}^{3}\right)$ was added to the resulting, ice-cooled solution. The mixture was then stirred at room temperature for 3 h . Then $10 \% \mathrm{aq}$. sodium carbonate was added and the mixture was extracted with methylene dichloride. The extract was washed, dried and evaporated to give a residue, which was purified by PLC (methylene dichloride-methanol, 98:2) to afford the benzazecine $31(5 \mathrm{mg}, 74 \%)$ as an oil, $v_{\text {max }} / \mathrm{cm}^{-1} 1730(\mathrm{OAc}) ; \delta_{\mathrm{H}}(200$ MHz) $7.33-7.12(4 \mathrm{H}, \mathrm{m}, \mathrm{ArH}), 6.81(1 \mathrm{H}, \mathrm{d}, J 16.5,8-\mathrm{H}), 5.30$ $(1 \mathrm{H}, \mathrm{d}, J 16.5,7-\mathrm{H}), 4.16\left(2 \mathrm{H}, \mathrm{br}\right.$ s, $\left.\mathrm{CH}_{2} \mathrm{OAc}\right), 2.97-2.68(2 \mathrm{H}$, $\left.\mathrm{m}, 1-\mathrm{H}_{2}\right), 2.38-2.05\left(4 \mathrm{H}, \mathrm{m}, 3-\mathrm{and} 5-\mathrm{H}_{2}\right), 2.36(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe})$, $2.09(3 \mathrm{H}, \mathrm{s}, \mathrm{OAc}), 1.78-1.53\left(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}_{2}\right)$ and $1.15(3 \mathrm{H}, \mathrm{s}$, 6-Me) (Found: $\mathrm{M}^{+}, 287.187 . \mathrm{C}_{18} \mathrm{H}_{25} \mathrm{NO}_{2}$ requires $\mathrm{M}, 287.188$ ).
(3ax,5ax,6ax,11bß,11cx)-( $\pm$ )-8-Benzoyl-5,5a, 6,6a,7,8,11b,11c,-octahydro-3a,5-dimethylfuro $[3,2-\mathrm{c}]$ indolo $[4,3-\mathrm{fg}]$ quinolin- 4 -(3a-$\mathrm{H})$-one 35.-Following the procedure given for compound 12, alkylation of the lithium enolate of compound $33^{3 a}(50 \mathrm{mg})$, prepared from diisopropylamine ( $0.092 \mathrm{~cm}^{3}$ ) and butyllithium ( $0.4 \mathrm{~cm}^{3} ; 1.59 \mathrm{~mol} \mathrm{dm}^{-3}$ solution in hexane) with methyl iodide ( $0.024 \mathrm{~cm}^{3}$ ), followed by purification of the crude product by a medium-pressure column chromatography (methylene dichlor-ide-methanol, $99: 1$ ), afforded the methyllactam 35 ( $46 \mathrm{mg}, 89 \%$ ) as crystals, m.p. $210-212^{\circ} \mathrm{C}$ (from methanol-ether); $v_{\text {max }} / \mathrm{cm}^{-1}$ $1638(\mathrm{NCO}) ; \delta_{\mathrm{H}}(200 \mathrm{MHz}) 7.70-6.94(8 \mathrm{H}, \mathrm{m}, \mathrm{ArH}), 6.39$ $(1 \mathrm{H}, \mathrm{d}, J 2.5,2-\mathrm{H}), 5.37(1 \mathrm{H}, \mathrm{d}, J 2.5,3-\mathrm{H}), 4.45(1 \mathrm{H}, \mathrm{d}, J 10.5$, $11 \mathrm{c}-\mathrm{H}: 24 \%$ intensity increase upon irradiation at $\delta 1.45$ ), 4.44 $\left(1 \mathrm{H}, \mathrm{br}, 7-\mathrm{H}^{\mathrm{B}}\right), 3.76\left(1 \mathrm{H}, \mathrm{t}, J 11.5,7-\mathrm{H}^{*}\right), 3.60(1 \mathrm{H}$, ddd, $J 12.5$, 10.5 and $3,5 \mathrm{a}-\mathrm{H}$ ), $3.40(1 \mathrm{H}, \mathrm{m}, 6 \mathrm{a}-\mathrm{H}), 3.08(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}), 3.00$ $(1 \mathrm{H}, \mathrm{t}, J 10.5,11 \mathrm{~b}-\mathrm{H}), 2.70\left(1 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}^{\mathrm{eq}}\right), 1.68(1 \mathrm{H}, \mathrm{br} \mathrm{q}, J 12$, $6-\mathrm{H}^{\mathrm{ax}}$ ) and $1.45(3 \mathrm{H}, \mathrm{s}, 3 \mathrm{a}-\mathrm{Me}: 2 \%$ intensity increase upon irradiation at $\delta 4.45$ ) (Found: C, 74.8; H, 6.1; N, 7.1. $\mathrm{C}_{25} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{3}$ requires $\mathrm{C}, 75.0 ; \mathrm{H}, 6.0 ; \mathrm{N}, 7.0 \%$ ).

Ozonolysis of the Methyl Lactam 35.-According to the procedure given for the preparation of compound $\mathbf{8}$, ozonolysis of the methyl lactam $35(1 \mathrm{~g})$ in methanol $\left(100 \mathrm{~cm}^{3}\right)$ in the presence of Oil Violet, followed by reduction with lithium aluminium hydride ( 1.5 g ) in anhydrous ether-THF ( $1: 1 ; 300$ $\mathrm{cm}^{3}$ ), gave a crystalline residue, which was recrystallised from methanol to afford ( $3 \beta, 8 \alpha, 9 \alpha$ )-( $\pm$ )-2,3-dihydro-9-hydroxy- 6,8 -dimethylergolin- 8 -ylmethanol 36 ( $384 \mathrm{mg}, 53 \%$ ) as crystals, m.p. $222-225{ }^{\circ} \mathrm{C}$ (decomp.); $v_{\max }(\mathrm{Nujol}) / \mathrm{cm}^{-1} 3276$ (NH); $\delta_{\mathrm{H}}[200$ $\left.\mathrm{MHz} ;\left(\mathrm{CD}_{3}\right)_{2} \mathrm{SO}\right] 7.39(1 \mathrm{H}, \mathrm{d}, J 8,12-\mathrm{H}), 6.82(1 \mathrm{H}, \mathrm{t}, J 8$, $13-\mathrm{H}), 6.36(1 \mathrm{H}, \mathrm{d}, J 8,14-\mathrm{H}), 3.72\left(2 \mathrm{H}\right.$, br s, $\mathrm{CH}_{2} \mathrm{OH}$ ), $3.52-$ $3.28\left(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}^{\mathrm{B}}\right.$ and $\left.9-\mathrm{H}^{\mathrm{eq}}\right), 3.04-2.84\left(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}^{\alpha}\right.$ and $3-\mathrm{H}), 2.84\left(1 \mathrm{H}, \mathrm{d}, J 11.5,7-\mathrm{H}^{\mathrm{eq}}\right), 2.80(1 \mathrm{H}, \mathrm{t}, J 10.5,10-\mathrm{H}), 2.40$
$\left(1 \mathrm{H}, \mathrm{brd}, J 11,4-\mathrm{H}^{\text {eq }}\right), 2.23(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}), 1.94(1 \mathrm{H}, \mathrm{brt}, J 10.5$, $5-\mathrm{H}), 1.80\left(1 \mathrm{H}, \mathrm{d}, J 11.5,7-\mathrm{H}^{\mathrm{ax}}\right), 1.20\left(1 \mathrm{H}, \mathrm{q}, J 11,4-\mathrm{H}^{\mathrm{ax}}\right), 0.92(3$ $\mathrm{H}, \mathrm{s}, 8-\mathrm{Me}$ ) (Found: $\mathrm{C}, 70.6 ; \mathrm{H}, 8.5 ; \mathrm{N}, 9.5 . \mathrm{C}_{17} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{2}$ requires $\mathrm{C}, 70.8 ; \mathrm{H}, 8.4 ; \mathrm{N}, 9.7 \%$ ).

Mother liquor was separated by PLC (methylene dichloridemethanol, 9:1) to give $(3 \beta, 8 \alpha, 9 \alpha)-( \pm)$-1-benzyl-2,3-dihydro-9-hydroxy-6,8-dimethylergolin-8-ylmethanol 37 ( $57 \mathrm{mg}, 6 \%$ ) as crystals, m.p. $196-198^{\circ} \mathrm{C}$ (from ethyl acetate); $\delta_{\mathrm{H}}(200 \mathrm{MHz})$ $7.54-7.20(6 \mathrm{H}, \mathrm{m}, 12-\mathrm{H}$ and ArH$), 7.06(1 \mathrm{H}, \mathrm{br} \mathrm{t}, J 8,13-\mathrm{H})$, $6.41(1 \mathrm{H}, \mathrm{d}, J 8,14-\mathrm{H}), 4.46$ and $3.91(2 \mathrm{H}, \mathrm{ABq}, J 14.5$, $\left.\mathrm{NCH}_{2} \mathrm{Ph}\right), 3.96(1 \mathrm{H}, \mathrm{br}$ dd, $J 11$ and $2, \mathrm{CH} \mathrm{HOH}), 3.88(1 \mathrm{H}$, d, $J 10, \mathrm{CH} H \mathrm{OH}), 3.66(1 \mathrm{H}, \mathrm{dd}, J 11$ and $8,9-\mathrm{H}), 3.50(1 \mathrm{H}, \mathrm{t}$, $\left.J 8,2-\mathrm{H}^{\mathrm{B}}\right), 3.37(1 \mathrm{H}$, br t, $J 11,10-\mathrm{H}), 3.15(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}), 2.93$ $\left(1 \mathrm{H}, \mathrm{d}, J 12,7-\mathrm{H}^{\mathrm{eq}}\right), 2.72\left(1 \mathrm{H}, \mathrm{dd}, J 12\right.$ and $\left.8,2-\mathrm{H}^{\alpha}\right), 2.58(1 \mathrm{H}$, d, $J 8,9-\mathrm{OH}), 2.40\left(1 \mathrm{H}\right.$, br ddd, $J 11,4.5$ and $\left.2,4-\mathrm{H}^{\text {eq }}\right), 2.29$ ( $3 \mathrm{H}, \mathrm{s}$, NMe), $2.28\left(1 \mathrm{H}\right.$, dd, $J 12$ and $\left.1.5,7-\mathrm{H}^{\mathrm{ax}}\right), 2.12(1 \mathrm{H}$, ddd, $J 11.5,10.5$ and $2,5-\mathrm{H}), 1.38\left(1 \mathrm{H}, \mathrm{q}, J 11.5,4-\mathrm{H}^{\mathrm{ax}}\right)$ and 0.95 ( $3 \mathrm{H}, \mathrm{s}, 8-\mathrm{Me}$ ) (Found: $\mathrm{M}^{+}, 378.231 . \mathrm{C}_{24} \mathrm{H}_{30} \mathrm{~N}_{2} \mathrm{O}_{2}$ requires M, 378.231).
$(3 \beta, 8 x, 9 x)-( \pm)-1-$ Acetyl-2,3-dihydro-9-hydroxy-6,8-dimethyl-ergolin-8-ylmethyl Acetate 38 .-Acetic anhydride $\left(0.32 \mathrm{~cm}^{3}\right)$ was added dropwise to a stirred, ice-cooled solution of the diol $\mathbf{3 6}(93 \mathrm{mg})$ in pyridine $\left(10 \mathrm{~cm}^{3}\right)$, and the mixture was stirred at room temperature for an additional 3 h . Then $10 \%$ aq. sodium carbonate was added and the mixture was extracted with methylene dichloride. The extract was washed, dried and evaporated to give a crystalline residue, which was recrystallised from ether-methylene dichloride to afford the acetate 38 $(112 \mathrm{mg}, 93 \%)$ as pale yellow crystals, m.p. $226-228.5^{\circ} \mathrm{C}$; $v_{\text {max }} / \mathrm{cm}^{-1} 1732(\mathrm{OAc})$ and $1650(\mathrm{NAc}) ; \delta_{\mathrm{H}}(200 \mathrm{MHz}) 7.90$ $(1 \mathrm{H}, \mathrm{d}, J 8,14-\mathrm{H}), 7.77(1 \mathrm{H}, \mathrm{d}, J 8,12-\mathrm{H}), 7.17(1 \mathrm{H}, \mathrm{t}, J 8$, $13-\mathrm{H}), 4.60$ and $4.42\left(2 \mathrm{H}, \mathrm{ABq}, J 11, \mathrm{CH}_{2} \mathrm{OAc}\right), 4.20(1 \mathrm{H}, \mathrm{t}, J$ $\left.9,2-\mathrm{H}^{\beta}\right), 3.61(1 \mathrm{H}$, br d$, J 11,9-\mathrm{H}), 3.60\left(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}^{\alpha}\right), 3.39$ $(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}), 2.99(1 \mathrm{H}$, br t$, J 10,10-\mathrm{H}), 2.91(1 \mathrm{H}, \mathrm{d}, J 12$, $7-\mathrm{H}^{\mathrm{eq}}$ ), $2.47\left(1 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}^{\mathrm{eq}}\right), 2.31(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}), 2.24(3 \mathrm{H}, \mathrm{s}$, NAc), $2.11(3 \mathrm{H}, \mathrm{s}, \mathrm{OAc}), 2.07(1 \mathrm{H}, \mathrm{t}, J 10,5-\mathrm{H}), 2.00(1 \mathrm{H}, \mathrm{d}, J$ $\left.12,7-\mathrm{H}^{\mathrm{ax}}\right), 1.41\left(1 \mathrm{H}\right.$, br q, $\left.J 12,4-\mathrm{H}^{\mathrm{ax}}\right)$ and $1.08(3 \mathrm{H}, \mathrm{s}, 8-\mathrm{Me})$ (Found: $\mathrm{C}, 66.55 ; \mathrm{H}, 7.4 ; \mathrm{N}, 7.5 . \mathrm{C}_{21} \mathrm{H}_{28} \mathrm{~N}_{2} \mathrm{O}_{4}-\frac{1}{10}-\mathrm{CH}_{2} \mathrm{Cl}_{2}$ requires $\mathrm{C}, 66.5 ; \mathrm{H}, 7.5 ; \mathrm{N}, 7.35 \%$ ).
$(3 a x, 5 a \beta, 6 a \beta, 11 b \beta, 11 c x)-( \pm)-8-$ Benzoyl-5,5a,6,6a,7,8,11b,11c-octahydro-3a,5-dimethylfuro[3,2-c]indolo[4,3-fg]quinolin-4(3a-$\mathrm{H})$-one 39.-Following the procedure given for compound $\mathbf{1 2}$, alkylation of the lithium enolate of compound $34(100 \mathrm{mg})$, prepared from diisopropylamine ( $0.18 \mathrm{~cm}^{3}$ ) and butyllithium ( $0.8 \mathrm{~cm}^{3} ; 1.59 \mathrm{~mol} \mathrm{dm}^{-3}$ solution in hexane), with methyl iodide ( $0.048 \mathrm{~cm}^{3}$ ), followed by purification of the crude product by medium-pressure column chromatography (methylene dichlor-ide-methanol, 99:1), afforded the methyl lactam 39 ( 98 mg , $95 \%$ ) as pale yellow crystals, m.p. 289-291 ${ }^{\circ} \mathrm{C}$ (decomp.) (from methanol-ether); $v_{\text {max }} / \mathrm{cm}^{-1} 1632 \quad(\mathrm{NCO}) ; \delta_{\mathrm{H}}(200 \mathrm{MHz})$ $7.64-7.00(8 \mathrm{H}, \mathrm{m}, \mathrm{ArH}), 6.32(1 \mathrm{H}, \mathrm{d}, J 2.5,2-\mathrm{H}), 5.33(1 \mathrm{H}, \mathrm{d}$, $J 2.5,3-\mathrm{H}), 4.34\left(1 \mathrm{H}, \mathrm{br}, 7-\mathrm{H}^{\beta}\right), 4.20(1 \mathrm{H}, \mathrm{d}, J 10.5,11 \mathrm{c}-\mathrm{H}), 3.74$ $\left(1 \mathrm{H}, \mathrm{t}, J 11.5,7-\mathrm{H}^{\alpha}\right), 3.63(1 \mathrm{H}, \mathrm{br}$ ddd, $J 12,6$ and $2.5,5 \mathrm{a}-\mathrm{H})$, $3.41(1 \mathrm{H}, \mathrm{m}, 6 \mathrm{a}-\mathrm{H}), 3.22(1 \mathrm{H}, \mathrm{dd}, J 10.5$ and $5.5,11 \mathrm{~b}-\mathrm{H}), 3.05$ $(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}), 2.38\left(1 \mathrm{H}\right.$, br d$\left., J 11,6-\mathrm{H}^{\mathrm{eq}}\right), 1.55(1 \mathrm{H}, \mathrm{q}, J 12$, $6-\mathrm{H}^{\mathrm{ax}}$ ) and 1.34 ( $3 \mathrm{H}, \mathrm{s}, 3 \mathrm{a}-\mathrm{Me}$ ) (Found: C, 74.7; H, 6.05; N, 6.9. $\mathrm{C}_{25} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{3}$ requires C, $75.0 ; \mathrm{H}, 6.0 ; \mathrm{N}, 7.0 \%$ ).

Ozonolysis of the Methyl Lactam 39 and Acetylation of the Product.-According to the procedure given for the preparation of compound 8 , ozonolysis of the methyl lactam $39(316 \mathrm{mg})$ in methanol ( $120 \mathrm{~cm}^{3}$ ) in the presence of Oil Violet, followed by reduction with lithium aluminium hydride ( 450 mg ) in anhydrous ether-THF $\left(1: 1 ; 100 \mathrm{~cm}^{3}\right)$, gave a crystalline residue, which was dissolved in pyridine ( $9 \mathrm{~cm}^{3}$ ). Then acetic anhydride ( $0.72 \mathrm{~cm}^{3}$ ) was added to this solution at $0{ }^{\circ} \mathrm{C}$ and the resulting
mixture was kept at room temperature for 15 h . After a similar work-up to that described for compound 9 , the residue was purified by medium-pressure column chromatography (methylene dichloride-methanol, 96:4) to give the acetate $40(207 \mathrm{mg}$, $70 \%$ and the $N$-benzyl derivative 41 ( $35 \mathrm{mg}, 11 \%$ ).
(3x,5x,8x,9x)-( $\pm)$-1-Acetyl-2,3-dihydro-9-hydroxy-6,8-di-methylergolin-8-ylmethyl acetate 40: pale yellow crystals, m.p. $220-222^{\circ} \mathrm{C}$ (from ethyl acetate); $v_{\text {max }} / \mathrm{cm}^{-1} 1730$ (OAc) and $1650(\mathrm{NAc}) ; \delta_{\mathbf{H}}(200 \mathrm{MHz}) 7.90(1 \mathrm{H}, \mathrm{d}, J 8,14-\mathrm{H}), 7.16(1 \mathrm{H}, \mathrm{t}$, $J 8,13-\mathrm{H}), 7.06(1 \mathrm{H}, \mathrm{br} \mathrm{d}, J 8,12-\mathrm{H}), 4.59$ and $4.44(2 \mathrm{H}, \mathrm{ABq}, J$ $\left.10, \mathrm{CH}_{2} \mathrm{OAc}\right), 4.20\left(1 \mathrm{H}, \mathrm{brt}, J 8.5,2-\mathrm{H}^{\beta}\right), 3.61(1 \mathrm{H}, \mathrm{dd}, J 12$ and $\left.9,2-\mathrm{H}^{x}\right), 3.42-3.20(3 \mathrm{H}, \mathrm{m}, 3-, 5-$, and $9-\mathrm{H}), 3.12(1 \mathrm{H}, \mathrm{dd}, J$ 11 and $5,10-\mathrm{H}), 2.58\left(1 \mathrm{H}, \mathrm{d}, J 12,7-\mathrm{H}^{\mathrm{eq}}\right), 2.43(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe})$, $2.33\left(1 \mathrm{H}, \mathrm{d}, J 12,7-\mathrm{H}^{\mathrm{ax}}\right), 2.25(3 \mathrm{H}, \mathrm{br}$ s, NAc), $2.13(1 \mathrm{H}, \mathrm{m}, 4-$ $\left.\mathrm{H}^{\mathrm{eq}}\right), 2.10(3 \mathrm{H}, \mathrm{s}, \mathrm{OAc}), 1.65\left(1 \mathrm{H}, \mathrm{br} \mathrm{q}, J 11.5,4-\mathrm{H}^{\mathrm{ax}}\right)$ and $1.00(3$ $\mathrm{H}, \mathrm{s}, 8-\mathrm{Me}$ ) (Found: $\mathrm{C}, 67.8 ; \mathrm{H}, 7.6 ; \mathrm{N}, 7.5 . \mathrm{C}_{21} \mathrm{H}_{28} \mathrm{~N}_{2} \mathrm{O}_{4}$ requires $\mathrm{C}, 67.7 ; \mathrm{H}, 7.6 ; \mathrm{N}, 7.5 \%$ ).
$(3 \alpha, 5 \alpha, 8 \alpha, 9 \alpha)-( \pm)$-1-Benzyl-2,3-dihydro-9-hydroxy-6,8-di-methylergolin-8-ylmethyl acetate 41: pale yellow crystals, m.p. 184-186 ${ }^{\circ} \mathrm{C}$ (from ethyl acetate); $v_{\max } / \mathrm{cm}^{-1} 1730$ (OAc); $\delta_{\mathrm{H}}(200 \mathrm{MHz}) 7.46-7.24(5 \mathrm{H}, \mathrm{m}, \mathrm{ArH}), 7.04(1 \mathrm{H}, \mathrm{t}, J 8,13-\mathrm{H})$, $6.72(1 \mathrm{H}, \mathrm{d}, J 8,12-\mathrm{H}), 6.40(1 \mathrm{H}, \mathrm{d}, J 8,14-\mathrm{H}), 4.61$ and 4.44 $\left(2 \mathrm{H}, \mathrm{ABq}, J 10, \mathrm{CH}_{2} \mathrm{OAc}\right), 4.48$ and $3.95(2 \mathrm{H}, \mathrm{ABq}, J 15$, $\left.\mathrm{NCH}_{2} \mathrm{Ph}\right), 3.53\left(1 \mathrm{H}, \mathrm{t}, J 7.5,2-\mathrm{H}^{\beta}\right), 3.35(1 \mathrm{H}, \mathrm{dd}, J 10.5$ and 4 , $9-\mathrm{H}), 3.31(1 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}), 3.07(1 \mathrm{H}, \mathrm{dd}, J 10.5$ and $5,10-\mathrm{H}), 3.06$ $(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}), 2.78\left(1 \mathrm{H}\right.$, br dd, $J 13$ and $\left.8,2-\mathrm{H}^{\alpha}\right), 2.57(1 \mathrm{H}, \mathrm{d}$, $\left.J 12,7-\mathrm{H}^{\mathrm{eq}}\right), 2.43(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}), 2.31\left(1 \mathrm{H}, \mathrm{d}, J 12,7-\mathrm{H}^{\mathrm{ax}}\right), 2.09$ $\left(1 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}^{\text {eq }}\right), 2.08(3 \mathrm{H}, \mathrm{s}, \mathrm{OAc}), 1.60\left(1 \mathrm{H}, \mathrm{q}, J 12,4-\mathrm{H}^{\mathrm{ax}}\right)$ and $1.00(3 \mathrm{H}, \mathrm{s}, 8-\mathrm{Me})$ (Found: C, 74.2; H, 7.7; N, 6.7. $\mathrm{C}_{26} \mathrm{H}_{32} \mathrm{~N}_{2} \mathrm{O}_{3}$ requires $\mathrm{C}, 74.25 ; \mathrm{H}, 7.7 ; \mathrm{N}, 6.7 \%$ ).

Fragmentation of the Acetate 38.-Mesyl chloride ( 20 mol equiv.) was added by portions to a solution of the acetate 38 in pyridine ( $0.11 \mathrm{~mol} \mathrm{dm}^{-3}$ ) at $0^{\circ} \mathrm{C}$ and the resulting solution was stirred at $20^{\circ} \mathrm{C}$ for 3 h . This reaction mixture was used for the next fragmentation reaction.
(a) The reaction mixture prepared from the acetate 38 (20 mg ) was warmed at $50^{\circ} \mathrm{C}$ for 4 h , and then $10 \%$ aq. ammonium hydroxide was added and the mixture was extracted with methylene dichloride. The organic layer was washed, dried and evaporated. The residue was purified by PLC (methylene dichloride-methanol, 92:8) to afford the allyl acetates 43 ( 2 mg , $11 \%$ ), $44(8 \mathrm{mg}, 43 \%)$, and $45(1 \mathrm{mg}, 5 \%)$.
$[2 a \alpha, 4 \beta, 5 \alpha(Z)]-( \pm)-3-\{1-$ Acetyl-1,2,2a,3,4,5-hexahydro-4-(methylamino)benz[cd]indol-5-yl\}-2-methylprop-2-enyl acetate 43: pale yellow crystals, m.p. $117-119.5^{\circ} \mathrm{C}$ (from acetone); $v_{\text {max }} / \mathrm{cm}^{-1} 1724(\mathrm{OAc})$ and $1656(\mathrm{NAc}) ; \delta_{\mathrm{H}}{ }^{*}(200 \mathrm{MHz}) 7.88$ $(1 \mathrm{H}, \mathrm{d}, J 8,14-\mathrm{H}), 7.18(1 \mathrm{H}, \mathrm{t}, J 8,13-\mathrm{H}), 6.70(1 \mathrm{H}, \mathrm{d}, J 8$, $12-\mathrm{H}), 5.37(1 \mathrm{H}, \mathrm{br} \mathrm{d}, J 10,9-\mathrm{H}), 4.96$ and $4.64(2 \mathrm{H}, \mathrm{ABq}, J 13$, $\left.\mathrm{CH}_{2} \mathrm{OAc}\right), 4.25\left(1 \mathrm{H}, \mathrm{t}, J 9,2-\mathrm{H}^{\beta}\right), 3.84(1 \mathrm{H}, \mathrm{t}, J 10,10-\mathrm{H}), 3.67$ ( 1 H, br t, $\left.J 10,2-\mathrm{H}^{\alpha}\right), 3.42(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}), 2.89(1 \mathrm{H}, \mathrm{br} \mathrm{t}, J 10$, $5-\mathrm{H}), 2.62(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}), 2.55\left(1 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}^{\mathrm{eq}}\right), 2.25(3 \mathrm{H}, \mathrm{s}, \mathrm{NAc})$, $2.11(3 \mathrm{H}, \mathrm{s}, \mathrm{OAc}), 1.89(3 \mathrm{H}, \mathrm{d}, J 1,8-\mathrm{Me})$ and $1.66(1 \mathrm{H}, \mathrm{br} \mathrm{q}$, $J 12,4-\mathrm{H}^{\mathrm{ax}}$ ) (Found: $\mathrm{C}, 69.95 ; \mathrm{H}, 7.6 ; \mathrm{N}, 8.0 . \mathrm{C}_{20} \mathrm{H}_{26} \mathrm{~N}_{2} \mathrm{O}_{3}$ requires $\mathrm{C}, 70.15 ; \mathrm{H}, 7.65 ; \mathrm{N}, 8.2 \%$ ).
[2ax, $4 \beta, 5 x(Z)]-( \pm)-3-\{1-$ Acetyl-1,2,2a,3,4,5-hexahydro-4-(N-methyl- N -methylsulfonylamino)benz[cd]indol-5-yl\}-2-methyl-prop-2-enyl acetate 44: pale yellow crystals, m.p. $170-171^{\circ} \mathrm{C}$ (from ethyl acetate); $v_{\text {max }} / \mathrm{cm}^{-1} 1732$ (OAc) and 1656 (NCO); $\delta_{\mathrm{H}}{ }^{*}(200 \mathrm{MHz}) 7.91(1 \mathrm{H}, \mathrm{d}, J 8,14-\mathrm{H}), 7.19(1 \mathrm{H}, \mathrm{t}, J 8,13-\mathrm{H})$, $6.74(1 \mathrm{H}, \mathrm{d}, J 8,12-\mathrm{H}), 5.43(1 \mathrm{H}, \mathrm{br}$ d, $J 10,9-\mathrm{H}), 4.73(2 \mathrm{H}$, br s, $\left.\mathrm{CH}_{2} \mathrm{OAc}\right), 4.30-4.08\left(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}^{\mathrm{B}}\right.$ and $\left.5-\mathrm{H}\right), 3.86(1 \mathrm{H}, \mathrm{t}, J 10$, $10-\mathrm{H}$ ), $3.70-3.36\left(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}^{\alpha}\right.$ and $3-\mathrm{H}$ ), 2.88 and 2.87 (each $3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}$ and NMs ), $2.25\left(1 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}^{\mathrm{eq}}\right), 2.25(3 \mathrm{H}, \mathrm{s}, \mathrm{NAc})$,

[^0]$2.13(3 \mathrm{H}, \mathrm{s}, \mathrm{OAc}), 1.90(3 \mathrm{H}, \mathrm{d}, J 1,8-\mathrm{Me})$ and $1.85(1 \mathrm{H}, \mathrm{m}$, 4- $\mathrm{H}^{\mathrm{ax}}$ ) (Found: $\mathrm{M}^{+}$, 420.172. $\mathrm{C}_{21} \mathrm{H}_{28} \mathrm{~N}_{2} \mathrm{O}_{5} \mathrm{~S}$ requires M , 420.172).
[2ax, $4 \beta, 5 \beta(\mathrm{E})]-( \pm)-3-\{1$-Acetyl-1,2,2a,3,4,5-hexahydro-4-(methylamino)benz[cd]indol-5-yl\}-2-methylprop-2-enyl acetate 45: a pale yellow oil, $v_{\max } / \mathrm{cm}^{-1} 1730$ (OAc) and 1656 (NAc); $\delta_{\mathrm{H}}{ }^{*}(200 \mathrm{MHz}) 7.87(1 \mathrm{H}$, br d, $J 8,14-\mathrm{H}), 7.16(1 \mathrm{H}, \mathrm{td}, J 8$ and $1,13-\mathrm{H}), 6.76(1 \mathrm{H}, \mathrm{d}, J 8,12-\mathrm{H}), 5.39(1 \mathrm{H}, \mathrm{dd}, J 11$ and $1.5,9-\mathrm{H}), 4.50\left(2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{OAc}\right), 4.25\left(1 \mathrm{H}, \mathrm{t}, J 8.5,2-\mathrm{H}^{\beta}\right), 4.17$ $(1 \mathrm{H}$, br dd, $J 11$ and $6,10-\mathrm{H}), 3.64\left(1 \mathrm{H}\right.$, br dd, $J 11$ and $\left.9,2-\mathrm{H}^{*}\right)$, $3.41(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}), 3.14(1 \mathrm{H}$, ddd, $J 12,6$ and $3,5-\mathrm{H}), 2.52(3 \mathrm{H}$, $\mathrm{s}, \mathrm{NMe}), 2.28\left(1 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}^{\mathrm{eq}}\right), 2.25(3 \mathrm{H}, \mathrm{s}, \mathrm{NAc}), 2.05(3 \mathrm{H}, \mathrm{s}$, OAc), 1.99 ( $3 \mathrm{H}, \mathrm{d}, J 1.5,8-\mathrm{Me}$ ) and $1.53\left(1 \mathrm{H}, \mathrm{q}, J 12,4-\mathrm{H}^{\mathrm{ax}}\right)$ (Found: $\mathrm{M}^{+}, 342.193 . \mathrm{C}_{20} \mathrm{H}_{26} \mathrm{~N}_{2} \mathrm{O}_{3}$ requires $\mathrm{M}, 342.194$ ).

The sulfonamide 44 was identical with a sample prepared by reaction of the allyl acetate 43 with mesyl chloride in pyridine upon comparison of their $R_{\mathrm{f}}$-values and IR and ${ }^{1} \mathrm{H}$ NMR spectra.
(b) Diethylamine ( $3 \mathrm{~cm}^{3}$ ) was added to the ice-cooled reaction mixture prepared from the acetate $38(20 \mathrm{mg})$ and the mixture was warmed at $50{ }^{\circ} \mathrm{C}$ for 4 h . Similar work-up as above gave two allyl acetates, 43 ( $3 \mathrm{mg}, 16 \%$ ) and $45(1 \mathrm{mg}, 5 \%)$, both of which were identical with the samples prepared above ( $a$ ) upon comparison of their $R_{\mathrm{f}}$-values and IR and ${ }^{1} \mathrm{H}$ NMR spectra.
(c) Diethyalmine ( $0.12 \mathrm{~cm}^{3}$ ) and ethanol ( $3 \mathrm{~cm}^{3}$ ) were added to the ice-cooled reaction mixture prepared from the acetate 38 ( 20 mg ) and the mixture was warmed at $50^{\circ} \mathrm{C}$ for 3 h . Similar work-up as above gave the allyl acetate $43(6 \mathrm{mg}, 33 \%)$, which was identical with the sample prepared above (a) upon comparison of their $R_{\mathrm{f}}$-values and IR and ${ }^{1} \mathrm{H}$ NMR spectra.
(d) Diethylamine $\left(0.52 \mathrm{~cm}^{3}\right)$ was added to the reaction mixture prepared from the acetate $38(78 \mathrm{mg})$, which was cooled to $-15^{\circ} \mathrm{C}$, and ethanol $\left(10 \mathrm{~cm}^{3}\right)$ was added to the mixture at $20^{\circ} \mathrm{C}$. The resulting mixture was warmed at $50^{\circ} \mathrm{C}$ for 3 h . Similar work-up as above gave the allyl acetate $43(40 \mathrm{mg}$, $56 \%$ ), which was identical with the sample prepared above ( $a$ ) upon comparison of their $R_{\mathrm{f}}$-values and IR and ${ }^{1} \mathrm{H}$ NMR spectra.

## (E)-( $8 \alpha, 12 a \beta)-( \pm)-2$-Acetyl-1,2,8,9,10,11,12,12a-octahydro-

 8,10-dimethylazecino [4,5-cd]indol-8-ylmethyl Acetate 46.-Mesyl chloride ( 0.16 ml ) was added to a stirred solution of the acetate $38(50 \mathrm{mg})$ in pyridine $\left(1.5 \mathrm{~cm}^{3}\right)$ at $0^{\circ} \mathrm{C}$, and the mixture was then stirred at room temperature for 3 h . Diethylamine $\left(0.31 \mathrm{~cm}^{3}\right)$ was added to this reaction mixture cooled to $-15^{\circ} \mathrm{C}$. After 20 min , ethanol ( $10 \mathrm{~cm}^{3}$ ) and sodium borohydride ( 50 mg ) were added to the mixture at $0^{\circ} \mathrm{C}$, and the mixture was then stirred at room temperature for 3 h . Water was added to the reaction mixture and the whole was extracted with methylene dichloride. The extract was washed, dried and evaporated. The residue was purified by PLC (methylene dichloride-methanol, 96:4) to give the benzazecine $46(28 \mathrm{mg}, 59 \%)$ as pale yellow crystals, m.p. $133.5-135.5^{\circ} \mathrm{C}$ (from methanol-ether); $v_{\text {max }} / \mathrm{cm}^{-1} 1730(\mathrm{OAc})$ and $1652(\mathrm{NAc}) ; \delta_{\mathrm{H}}(200 \mathrm{MHz}) 8.10$ $(1 \mathrm{H}, \mathrm{d}, J 8,5-\mathrm{H}), 7.21(1 \mathrm{H}, \mathrm{t}, J 8,4-\mathrm{H}), 6.95(1 \mathrm{H}, \mathrm{d}, J 8,3-\mathrm{H})$, $6.67(1 \mathrm{H}, \mathrm{d}, J 17,6-\mathrm{H}), 5.71(1 \mathrm{H}, \mathrm{d}, J 17,7-\mathrm{H}), 4.19(1 \mathrm{H}, \mathrm{t}, J 9$, $\left.1-\mathrm{H}^{\beta}\right), 4.06$ and $4.04\left(2 \mathrm{H}, \mathrm{ABq}, J 13, \mathrm{CH}_{2} \mathrm{OAc}\right), 3.80(1 \mathrm{H}, \mathrm{m}$, $12 \mathrm{a}-\mathrm{H}), 3.65\left(1 \mathrm{H}\right.$, br d, $\left.J 10,1-\mathrm{H}^{\alpha}\right), 2.52-2.02(4 \mathrm{H}, \mathrm{m}, 9$ - and 11$\mathrm{H}^{2}$ ), 2.44 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}$ ), $2.24(3 \mathrm{H}, \mathrm{s}, \mathrm{NAc}), 2.10(3 \mathrm{H}, \mathrm{s}, \mathrm{OAc})$, 1.92-1.49 ( $2 \mathrm{H}, \mathrm{m}, 12-\mathrm{H}_{2}$ ) and $1.25(3 \mathrm{H}, \mathrm{s}, 8-\mathrm{Me})$ (Found: $\mathrm{M}^{+}$, 356.208. $\mathrm{C}_{21} \mathrm{H}_{28} \mathrm{~N}_{2} \mathrm{O}_{3}$ requires $\mathrm{M}, 356.210$ ).Fragmentation of the Acetate 40 .-Mesyl chloride ( $0.1 \mathrm{~cm}^{3}$ ) was added to a solution of the acetate $\mathbf{4 0}(20 \mathrm{mg})$ in pyridine

[^1]$\left(0.5 \mathrm{~cm}^{3}\right)$ at $0{ }^{\circ} \mathrm{C}$ and the resulting solution was stirred at room temperature for 3 h . Then $10 \%$ aq. ammonium hydroxide was added, and the mixture was extracted with methylene dichloride. The extract was washed, dried and evaporated to give a residue, which was, without purification, dissolved in either absolute ethanol ( $3 \mathrm{~cm}^{3}$ ), DMSO ( $0.5 \mathrm{~cm}^{3}$ ) or ethylene glycol $\left(2 \mathrm{~cm}^{3}\right)$. The mixture was treated at the temperature and for the time described in the relevant entry in Table 2. After similar work-up as given for the fragmentation of compound 16, the crude product was separated by PLC (methylene dichloridemethanol, $92: 8$ ) to afford the allyl acetates 48 and 49 in the isolated yields shown in Table 2.
[2ax,4 $3,5 x(\mathrm{E})]-( \pm)-3-\{1-$ Acetyl-1,2,2a,3,4,5-hexahydro-4-(methylamino)benz[cd]indol-5-yl\}-2-methylprop-2-enyl acetate 48: pale yellow crystals, m.p. $79-82^{\circ} \mathrm{C}$ (from acetone); $v_{\text {max }} / \mathrm{cm}^{-1} 1732(\mathrm{OAc})$ and $1652(\mathrm{NAc}) ; \delta_{\mathrm{H}}{ }^{*}(200 \mathrm{MHz}) 7.88$ $(1 \mathrm{H}, \mathrm{d}, J 8,14-\mathrm{H}), 7.18(1 \mathrm{H}, \mathrm{t}, J 8,13-\mathrm{H}), 6.73(1 \mathrm{H}, \mathrm{d}, J 8$, $12-\mathrm{H}), 5.46(1 \mathrm{H}, \mathrm{br}$ d, $J 10,9-\mathrm{H}), 4.64\left(2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{OAc}\right), 4.25$ $\left(1 \mathrm{H}, \mathrm{brt}, J 10,2-\mathrm{H}^{\beta}\right), 3.66\left(1 \mathrm{H}\right.$, br t, $\left.J 10.5,2-\mathrm{H}^{\alpha}\right), 3.59(1 \mathrm{H}, \mathrm{br}$ t, $J 10,10-\mathrm{H}), 3.43(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}), 2.70(1 \mathrm{H}$, br ddd, $J 12,10$ and $3,5-\mathrm{H}), 2.53\left(1 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}^{\mathrm{eq}}\right), 2.50(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}), 2.26(3 \mathrm{H}, \mathrm{s}$, NAc), $2.11(3 \mathrm{H}, \mathrm{s}, \mathrm{OAc}), 1.89(3 \mathrm{H}, \mathrm{d}, J 1,8-\mathrm{Me})$ and $1.39(1 \mathrm{H}$, q, $J 12,4-\mathrm{H}^{\text {ax }}$ ) (Found: $\mathrm{M}^{+}, 342.195 . \mathrm{C}_{20} \mathrm{H}_{26} \mathrm{~N}_{2} \mathrm{O}_{3}$ requires M , 342.194).
$[2 a \alpha, 4 \beta, 5 \beta(Z)]-( \pm)-3-\{1-$ Acetyl-1,2,2a,3,4,5-hexahydro-4-(methylamino)benz[cd]indol-5-yl\}-2-methylprop-2-enyl acetate 49: pale yellow crystals, m.p. $144-146^{\circ} \mathrm{C}$ (from ethyl acetate); $v_{\text {max }} / \mathrm{cm}^{-1} 1730(\mathrm{OAc})$ and $1654(\mathrm{NAc}) ; \delta_{\mathrm{H}}{ }^{*}(200 \mathrm{MHz}) 7.86$ $(1 \mathrm{H}, \mathrm{d}, J 8,14-\mathrm{H}), 7.15(1 \mathrm{H}, \mathrm{t}, J 8,13-\mathrm{H}), 6.71(1 \mathrm{H}, \mathrm{d}, J 8$, $12-\mathrm{H}), 5.32(1 \mathrm{H}$, br d, $J 11,9-\mathrm{H}), 4.98$ and $4.76(2 \mathrm{H}, \mathrm{ABq}, J 12$, $\left.\mathrm{CH}_{2} \mathrm{OAc}\right), 4.24\left(1 \mathrm{H}, \mathrm{t}, J 8.5,2-\mathrm{H}^{\beta}\right), 4.17(1 \mathrm{H}$, dd, J 11 and 6 , $10-\mathrm{H}), 3.64\left(1 \mathrm{H}\right.$, br dd, $J 11$ and $\left.9,2-\mathrm{H}^{*}\right), 3.38(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H})$, $3.06(1 \mathrm{H}$, br ddd, $J 11.5,6$ and $3,5-\mathrm{H}), 2.53(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}), 2.26$ ( $1 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}^{\mathrm{eq}}$ ), $2.24(3 \mathrm{H}, \mathrm{s}, \mathrm{NAc}), 2.12(3 \mathrm{H}, \mathrm{s}, \mathrm{OAc}), 1.79(3 \mathrm{H}$, br s, $8-\mathrm{Me}$ ) and $1.53\left(1 \mathrm{H}, \mathrm{q}, J 12,4-\mathrm{H}^{\mathrm{ax}}\right)$ (Found: $\mathrm{M}^{+}, 342.194$ ).
( $\pm$ )-Isochanoclavine-I 50.—A solution of compound 43 (26.5 mg ) in a mixture of conc. hydrochloric acid ( $0.38 \mathrm{~cm}^{3}$ ) and methanol ( $5 \mathrm{~cm}^{3}$ ) was warmed at $80^{\circ} \mathrm{C}$ for 3 h . Removal of the solvent gave a residue, which was dissolved in anhydrous THF ( $8 \mathrm{~cm}^{3}$ ), and triethylamine $\left(0.1 \mathrm{~cm}^{3}\right.$ ), indole ( 27.6 mg ), and benzeneseleninic anhydride ( 21.2 mg ) were added in turn to this solution at room temperature. The resulting mixture was warmed at $50^{\circ} \mathrm{C}$ for 2 h , the solvent was partly evaporated off, $10 \%$ aq. sodium carbonate was added to the residue, and the mixture was extracted with methylene dichloride. The extract was washed, dried and evaporated to give a residue, which was purified by PLC (chloroform-methanol- $28 \%$ aq. ammonium hydroxide, $9: 1: 0.02$ ) to afford ( $\pm$ )-isochanoclavine-I 50 (16.4 $\mathrm{mg}, 83 \%$ ) as pale brown crystals, m.p. $190-192{ }^{\circ} \mathrm{C}$ (decomp.) (from aq. methanol) [lit., ${ }^{14} 200-201{ }^{\circ} \mathrm{C}$ decomp.)]; $\delta_{\mathrm{H}}{ }^{*}(200$ $\mathrm{mHz} ;\left[{ }^{2} \mathrm{H}_{5}\right]$ pyridine $) 11.64(1 \mathrm{H}$, br s, $1-\mathrm{H}), 7.45(1 \mathrm{H}, \mathrm{d}, J 7$, 12 - or $14-\mathrm{H}), 7.40-7.20(2 \mathrm{H}, \mathrm{m}, 2-$ and $13-\mathrm{H}), 7.09(1 \mathrm{H}, \mathrm{d}, J 7$, 14 - or $12-\mathrm{H}), 5.53(1 \mathrm{H}$, br d, $J 10,9-\mathrm{H}), 4.74$ and $4.60(2 \mathrm{H}, \mathrm{ABq}$, $\left.J 12, \mathrm{CH}_{2} \mathrm{OH}\right), 4.30(1 \mathrm{H}$, br dd, $J 10$ and $8,10-\mathrm{H}), 3.44(1 \mathrm{H}$, dd, $J 14$ and $\left.3,4-\mathrm{H}^{\mathrm{eq}}\right), 3.00-2.89\left(2 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}^{\mathrm{ax}}\right.$ and $\left.5-\mathrm{H}\right) 2.44(3 \mathrm{H}$, s , NMe) and 2.19 ( $3 \mathrm{H}, \mathrm{d}, J 1,8-\mathrm{Me}$ ) (Found: C, 73.5; H, 7.75; N, 10.8. Calc. for $\mathrm{C}_{16} \mathrm{H}_{20} \mathrm{~N}_{2} \mathrm{O}: \mathrm{C}, 73.7 ; \mathrm{H}, 7.9 ; \mathrm{N}, 10.7 \%$ ). IR and ${ }^{1} \mathrm{H}$ NMR spectra and $R_{\mathrm{f}}$-value of compound $( \pm)$ - 50 were found to be identical with those of $( \pm)$-isochanoclavine-I donated by Professor Somei.
( $\pm$ )-Chanoclavine-I 51.-According to the procedure given for compound 50 , hydrolysis of compound $48(30 \mathrm{mg})$ in methanol ( $6 \mathrm{~cm}^{3}$ ) containing conc. hydrochloric acid $\left(0.5 \mathrm{~cm}^{3}\right)$ gave a crude product. Without purification, a mixture of the crude product, triethylamine ( $0.06 \mathrm{~cm}^{3}$ ), indole ( 31 mg ), benzene seleninic anhydride ( 20 mg ), and THF ( $9 \mathrm{~cm}^{3}$ ) was
heated at $40^{\circ} \mathrm{C}$ for 3 h . Work-up as given for the preparation of compound 50 gave ( $\pm$ )-chanoclavine-I $51(15 \mathrm{mg}, 67 \%)$ as pale yellow crystals, m.p. 192-194 ${ }^{\circ} \mathrm{C}$ (decomp.) (from acetone) [lit., ${ }^{14}$ 194-195 ${ }^{\circ} \mathrm{C}$ (decomp.)]; $\delta_{\mathrm{H}}{ }^{*}\left(200 \mathrm{MHz} ; \quad\left[{ }^{2} \mathrm{H}_{5}\right]\right.$ pyridine) $11.54(1 \mathrm{H}$, br s, 1-H), $7.42(1 \mathrm{H}, \mathrm{d}, J 8,12-$ or $14-\mathrm{H}), 7.38-$ $7.18(2 \mathrm{H}, \mathrm{m}, 2$ - and $13-\mathrm{H}), 7.03(1 \mathrm{H}, \mathrm{d}, J 8,14$ - or $12-\mathrm{H}), 5.89$ ( 1 H, br d, $J 10,9-\mathrm{H}), 4.46\left(2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{OH}\right), 4.21(1 \mathrm{H}, \mathrm{dd}, J 10$ and $8,10-\mathrm{H}), 3.42\left(1 \mathrm{H}, \mathrm{dd}, J 14\right.$ and $\left.3,4-\mathrm{H}^{\mathrm{eq}}\right), 3.02(1 \mathrm{H}$, br td, $J 9$ and $3,5-\mathrm{H}), 2.89\left(1 \mathrm{H}\right.$, dd, $J 14$ and $\left.9,4-\mathrm{H}^{\mathrm{ax}}\right), 2.40(3 \mathrm{H}, \mathrm{s}$, NMe ) and 2.03 ( 3 H , br s, 8-Me) (Found: $\mathrm{M}^{+}$, 256.158. Calc. for $\mathrm{C}_{16} \mathrm{H}_{20} \mathrm{~N}_{2} \mathrm{O}: M, 256.158$ ). IR and ${ }^{1} \mathrm{H}$ NMR spectra and the $R_{\mathrm{f}}$-value of the product ( $\pm$ )-51 were found to be identical with those of $( \pm)$-chanoclavine-I supplied by Professor Somei.

* Numbering scheme for compounds $\mathbf{4 3} \mathbf{4 5}, 48$ and 49-51 follows the ergoline system.


## Acknowledgements

We are grateful to Professor M. Somei, Kanazawa University (Japan), for providing authentic samples of $( \pm$ )-chanoclavine-I and ( $\pm$ )-isochanoclavine-I and their NMR and IR spectra, and also for generous discussions on the technical problems encountered.

2 Preliminary communications: T. Kiguchi, T. Naito and I. Ninomiya, Heterocycles, 1987, 26, 1747; T. Kiguchi, N. Kuninobu, T. Naito and I. Ninomiya, Heterocycles, 1989, 28, 19.

3 (a) I. Ninomiya, C. Hashimoto, T. Kiguchi and T. Naito, J. Chem. Soc., Perkin Trans. I, 1985, 941; (b) I. Ninomiya, C. Hashimoto, T. Kiguchi, T. Naito, D. H. R. Barton, X. Lusinchi and P. Milliet, J. Chem. Soc., Perkin Trans. 1, 1990, 707; (c) I. Ninomiya, T. Kiguchi, C. Hashimoto and T. Naito, Chem. Pharm. Bull., 1991, 39, 23.

4 1. Ninomiya and T. Kiguchi, in The Alkaloids, ed. A. Brossi, Academic, New York, 1990, vol. 38, pp. 1-156.
5 C. A. Grob, Angew. Chem., Int. Ed. Engl., 1969, 8, 535.
6 C. A. Grob, H. R. Kiefer, H. J. Lutz and H. J. Wilkens, Helv. Chim. Acta, 1967, 50, 416; J. A. Marshall and J. H. Babler, J. Org. Chem., 1969, 34, 4186.
7 I. Ninomiya, C. Hashimoto, T. Kiguchi and T. Naito, (a) J. Chem. Soc., Perkin Trans. 1, 1984, 2911; (b) Chem. Pharm. Bull., 1986, 34, 2799.

8 H. M. Bell, C. W. Vanderslice and A. Spehar, J. Org. Chem., 1969, 34, 3923.

9 C. A. Grob, W. Kunz and P. R. Marbet, Tetrahedron Lell., 1975, 2613.
10 W. Oppolzer, J. I. Grayson, H. Wegmann and M. Urrea, Tetrahedron, 1983, 39, 3695.
11 D. Stauffacher and H. Tscherter, Helv. Chim. Acta, 1964, 47, 2186.
12 M. Somei, F. Yamada and Y. Makita, Heterocycles, 1987, 26, 895.
13 R. Ramage, V. W. Armstrong and S. Coulton, Tetrahedron, 1981, 37, Suppl. 1, 157; J. Rebek, Jr., D. F. Tai and Y. K. Shue, J. Am. Chem. Soc., 1984, 106, 1813.
14 M. Somei, Y. Makita and F. Yamada, Chem. Pharm. Bull., 1986, 34, 94.

15 T. Veysoglu, L. A. Mitscher and J. K. Swayze, Synthesis, 1980, 807.


[^0]:    * Numbering scheme for compounds $\mathbf{4 3 - 4 5}, 48$ and 49-51 follows the ergoline system.

[^1]:    * Numbering scheme for compounds 43-45, 48 and 49-51 follows the ergoline system.

